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INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<p>(51) International Patent Classification⁶ : A61K 7/13, 7/06</p>	<p>A1</p>	<p>(11) International Publication Number: WO 99/13845 (43) International Publication Date: 25 March 1999 (25.03.99)</p>
<p>(21) International Application Number: PCT/US97/16410 (22) International Filing Date: 17 September 1997 (17.09.97) (71) Applicant (for all designated States except US): THE PROCTER & GAMBLE COMPANY [US/US]; One Procter & Gamble Plaza, Cincinnati, OH 45202 (US). (72) Inventors; and (75) Inventors/Applicants (for US only): LUO, Xiaochun [CN/JP]; 5-15-703, Koyo-cho Naka, Higashinada-ku, Kobe 658 (JP). TIAN, Minmin [CN/JP]; #203, 4-2-41, Uozaki-nishimachi, Higashinada-ku, Kobe 658 (JP). MAO, Hsiang, Kuen [US/JP]; 2-23-13, Sumiyoshi-honmachi, Higashinada-ku, Kobe 658 (JP). (74) Agents: REED, T., David et al.; The Procter & Gamble Company, 5299 Spring Grove Avenue, Cincinnati, OH 45217-1087 (US).</p>		<p>(81) Designated States: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, HU, ID, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, ARIPO patent (GH, KE, LS, MW, SD, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG).</p> <p>Published <i>With international search report.</i></p>
<p>(54) Title: HAIR CARE COMPOSITIONS COMPRISING OPTICAL BRIGHTENERS WHICH ALTER HAIR COLORS (57) Abstract <p>Disclosed are hair care compositions comprising: (a) an optical brightener in an amount which alters the color of the hair to which the composition is applied by from about 0.2 units to about 15 units of either or both of a-value and a b-value; and (b) a hair care active selected from the group consisting of a detergent surfactant, a conditioning agent, a fixative polymer, and mixtures thereof. Further disclosed are methods of altering the color of the hair by applying an optical brightener to the hair.</p></p>		

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HAIR CARE COMPOSITIONS COMPRISING OPTICAL BRIGHTENERS WHICH ALTER HAIR COLORS

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TECHNICAL FIELD

The present invention relates to a hair care composition comprising optical brighteners which alter the color of hair.

BACKGROUND

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The desire to regain the natural color and shine of damaged hair and the desire to alter the color of the hair to be more appealing are widely held. Damaged hair is perceived by the consumer as unfavorable appearances and less manageability of the hair. Such unfavorable appearances include alteration and fading of original color, less shine, and less luster.

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A common way for alleviating the unfavorable appearances of damaged hair and to achieve appealing hair color is to dye the hair to the color desired. Dyeing the hair would provide the consumer with a stable color of hair for a relatively long period. However, dyeing the hair is generally time-consuming, cumbersome, and messy. Dyestuff may also be chemically harsh to the hair, scalp, and skin. The hair can be further damaged by dyeing. Thus, hair dye products are not suitable for daily use. Further, dyeing can leave the hair with a dull appearance, making the hair look less shiny.

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Based on the foregoing, there is a need for a hair composition which can be used daily and which can alter the color of the hair, while enhancing the shininess of the hair, and protecting the hair from further damage.

35

Use of optical brighteners, or compounds otherwise described by names such as fluorescent whitening agents, fluorescent brighteners, or fluorescent dyes, in the hair care field has been known in the art, such as in United States Patent 3,658,985, United States Patent 4,312,855, Canadian Patent 1,255,603, United States Patent 3,577,528, Great Britain Patent Specification 1,328,108,

South African Application 676,049, European Publication 87,060, and Great Britain Patent Specification 2,307,639.

5 The color alteration of the hair can be technically measured by L,a,b color parameters. When either or both of the a-value and b-value are altered to a certain degree, one perceives that the appearance of the hair is altered in such a way that the unfavorable appearances of damaged hair is alleviated, or that the hair color has become more appealing. There is a need for a hair composition which can be used daily and which can provide perceivable degree of color alteration.

10 None of the existing art provides all of the advantages and benefits of the present invention.

SUMMARY

15 The present invention is directed to a hair care composition comprising: (a) an optical brightener in an amount which alters the color of the hair to which the composition is applied by from about 0.2 units to about 15 units of either or both of a-value and b-value; and (b) a hair care active selected from the group consisting of a deterative surfactant, a conditioning agent, a fixative polymer, and mixtures thereof.

20 The present invention is further directed to a method of altering the color of the hair by applying an optical brightener to the hair.

These and other features, aspects, and advantages of the present invention will become evident to those skilled in the art from a reading of the present disclosure.

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BRIEF DESCRIPTION OF THE DRAWING

While the specification concludes with claims particularly pointing out and distinctly claiming the invention, it is believed that the present invention will be better understood from the following description in conjunction with Figure 1 which describes the "L,a,b, Color Coordinate" showing the color parameters.

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DETAILED DESCRIPTION

While the specification concludes with claims particularly pointing and distinctly claiming the invention, it is believed that the present invention will be better understood from the following description.

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All percentages are by weight of the total composition unless otherwise indicated. All ratios are weight ratios unless otherwise indicated. All percentages, ratios, and levels of ingredients referred to herein are based on the actual amount of the ingredient, and do not include solvents, fillers, or other materials with which the ingredient may be combined as commercially available products, unless otherwise indicated.

As used herein, "comprising" means that other steps and other ingredients which do not affect the end result can be added. This term encompasses the terms "consisting of" and "consisting essentially of".

All cited references are incorporated herein by reference in their entireties. Citation of any reference is not an admission regarding any determination as to its availability as prior art to the claimed invention.

OPTICAL BRIGHTENERS

Optical brighteners are compounds which absorb ultraviolet light and re-emit the energy in the form of visible light. Specifically, the optical brighteners useful herein have an absorption, preferably a major absorption peak, between a wavelength of about 1nm and about 420nm, and an emission, preferably a major emission peak, between a wavelength of about 360nm and about 830nm; wherein the major absorption peak has a shorter wavelength than the major emission peak. More preferably, the optical brighteners useful herein have a major absorption peak between a wavelength of about 200nm and about 420nm, and a major emission peak between a wavelength of about 400nm and about 780nm. Optical brighteners may or may not have minor absorption peaks in the visible range between a wavelength of about 360nm and about 830nm. Optical brighteners can be described by other names in the art and in other industries, such as fluorescent whitening agents, fluorescent brighteners, and fluorescent dyes.

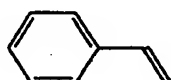
When applied to hair via suitable vehicles, optical brighteners herein provide benefits to the hair in three areas. First, optical brighteners herein alter the color of the hair by emitting light in the visible range. Second, optical brighteners herein enhance the shine of the hair by emitting light in the visible range. Third, optical brighteners herein protect the hair from ultraviolet light by absorbing ultraviolet light.

Optical brighteners in general are based on the structures of aromatic and heteroaromatic systems which provide these unique characteristics. The optical brighteners useful in the present invention can be classified according to their

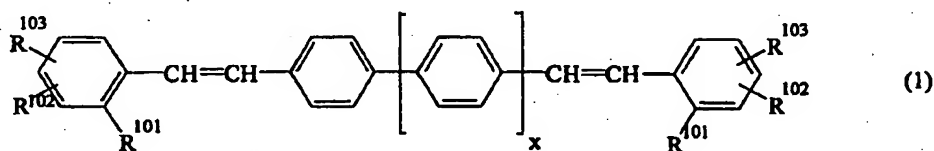
base structures, as described hereafter. Preferable optical brighteners herein include polystyrylstilbenes, triazinstilbenes, hydroxycoumarins, aminocoumarins, triazoles, pyrazolines, oxazoles, pyrenes, porphyrins, and imidazoles.

Polystyrylstilbenes

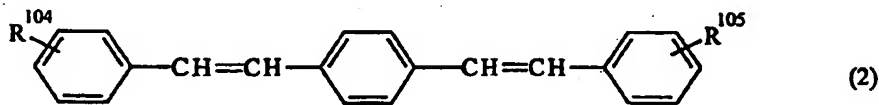
- 5 Polystyrylstilbenes are a class of compounds having two or more of the following base structure:



- 10 Polystyrylstilbenes useful in the present invention include those having formulae (1), (2) and (3):

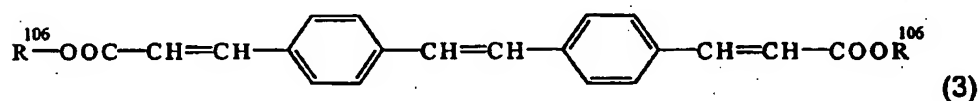


- 15 wherein R^{101} is H, OH, SO_3M , $COOM$, OSO_3M , $OPO(OH)OM$, wherein M is H, Na, K, Ca, Mg, ammonium, mono-, di-, tri- or tetra- C_1 - C_{30} -alkylammonium, mono-, di- or tri- C_1 - C_{30} -hydroxyalkylammonium or ammonium that is di- or tri-substituted with by a mixture of C_1 - C_{30} -alkyl and C_1 - C_{30} -hydroxyalkyl groups; or $SO_2N(C_1-C_{30}-alkyl)_2$, $O-(C_1-C_{30}-alkyl)$, CN, Cl, $COO(C_1-C_{30}-alkyl)$, $CON(C_1-C_{30}-alkyl)_2$ or $O(CH_2)_3N^+(CH_3)_2X^-$ wherein X^- is an anion of a chloride, bromide, iodide, formate, acetate, propionate, glycolate, lactate, acrylate, methanephosphonate, phosphite, dimethyl or diethyl phosphite anion; CN, or alkyl of 1 to 30 carbons, R^{102} and R^{103} , independently, are H, SO_3M wherein M is as previously defined; and x is 0 or 1; wherein the compound has a trans-coplanar orientation or cis-coplanar orientation; preferably x is 1, R^{101} is SO_3Na and R^{102} and R^{103} are H; wherein the compound has a trans-coplanar orientation;
- 20
- 25
- 30



wherein R^{104} and R^{105} , independently, are CN, $\text{COO}(\text{C}_1\text{-C}_{30}\text{-alkyl})$, $\text{CONHC}_1\text{-C}_4\text{-alkyl}$, or $\text{CON}(\text{C}_1\text{-C}_4\text{-alkyl})_2$, wherein the compound has a trans-coplanar orientation or cis-coplanar orientation; preferably R^{104} and R^{105} is 2-cyano, wherein the compound has a trans-coplanar orientation; and

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wherein each R^{106} , independently, is H, or alkyl of 1 to 30 carbons; and wherein the compound has a trans-coplanar orientation or cis-coplanar orientation, preferably a trans-coplanar orientation.

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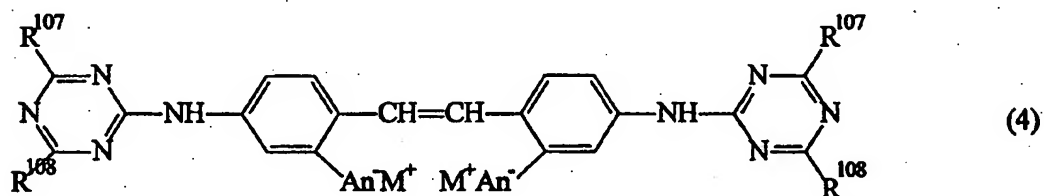
Suitable polystyrylstilbenes include disodium-1, 4'-bis(2-sulfoethyl) bisphenyl (C.I. Fluorescent Brightener 351) with tradename Tinopal CBS-X available from Ciba Specialty Chemicals, 1,4-bis(2-cyanostyryl)benzene (C.I. Fluorescent Brightener 199) with tradename Ultraphor RN available from BASF.

15 Triazinstilbenes

Triazinstilbenes are a class of compounds having both triazin and stilbene structures in the same molecule.

Triazinstilbenes useful in the present invention include those having formulae (4):

20



wherein R^{107} and R^{108} , independently, are phenylamino, mono- or disulfonated phenylamino, morpholino, $\text{N}(\text{CH}_2\text{CH}_2\text{OH})_2$, $\text{N}(\text{CH}_3)(\text{CH}_2\text{CH}_2\text{OH})$, NH_2 , $\text{N}(\text{C}_1\text{-C}_4\text{-alkyl})_2$, OCH_3 , Cl, $\text{NH}(\text{CH}_2)_{1-4}\text{SO}_3\text{H}$ or $\text{NH}(\text{CH}_2)_{1-4}\text{OH}$; An^- is an anion of a carboxylate, sulfate, sulfonate, or phosphate, and M is as previously defined, wherein the compound has a trans-coplanar orientation or cis-coplanar orientation; preferably R^{107} is 2, 5-disulfophenylamino and each R^{108} is morpholino; or each R^{107} is 2, 5-disulfophenylamino and each R^{108} is $\text{N}(\text{C}_2\text{H}_5)_2$; or each R^{107} is 3-sulfophenyl and each R^{108} is $\text{NH}(\text{CH}_2\text{CH}_2\text{OH})$ or $\text{N}(\text{CH}_2\text{CH}_2\text{OH})_2$; or each R^{107} is 4-sulfophenyl and each R^{108} is

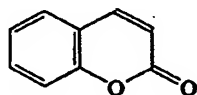
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$N(CH_2CH_2OH)_2$; and in each case, the sulfo group is SO_3M in which M is sodium; wherein the compound has a trans-coplanar orientation.

Suitable triazinstilbenes include 4,4'-bis-[(4-anilino-6-bis(2-hydroxyethyl)amino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disulfonic acid with tradename Tinopal UNPA-GX available from Ciba Specialty Chemicals, 4,4'-bis-[(4-anilino-6-morpholine-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disodium sulfonate with tradename Tinopal AMS-GX available from Ciba Specialty Chemicals, 4,4'-bis-[(4-anilino-6-(2-hydroxyethyl)methyl amino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disodium sulfonate with tradename Tinopal 5BM-GX available from Ciba Specialty Chemicals, 4,4'-bis-[(4,6-dianilino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disodium sulfonate, 4,4'-bis-[(4-anilino-6-methylamino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disodium sulfonate, 4,4'-bis-[(4-anilino-6-ethylamino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disodium sulfonate, and 4,4'-bis(4-phenyl-1,2,3-triazol-2-yl)stilbene-2,2'-disulfonic acid.

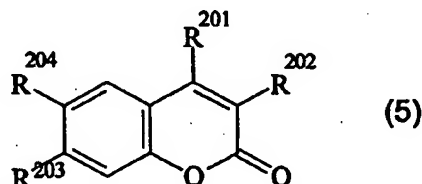
15 Hydroxycoumarins

Hydroxycoumarins are a class of compounds having the following base coumarin structure and having at least one hydroxy moiety:



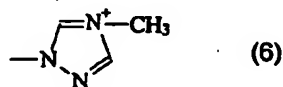
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Hydroxycoumarins useful in the present invention include those having formulae (5):



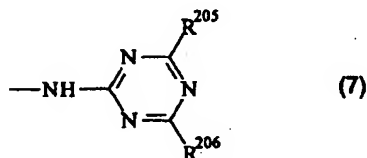
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wherein R^{201} is H, OH, Cl, CH_3 , CH_2COOH , CH_2SO_3H , CH_2OSO_3H , or $CH_2OPO(OH)OH$, R^{202} is H, phenyl, $COO-C_1-C_{30}$ -alkyl, glucose, or a group of formula (6):



30

and R²⁰³ is OH, or O-C₁-C₃₀-alkyl, and R²⁰⁴ is OH or O-C₁-C₃₀ alkyl, glycoside, or a group of the following formula (7):



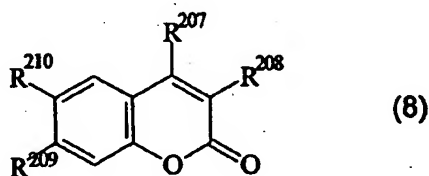
wherein R²⁰⁵ and R²⁰⁶ are independently, phenylamino, mono- or disulfonated phenylamino, morpholino, N(CH₂CH₂OH)₂, N(CH₃)(CH₂CH₂OH), NH₂, N(C₁-C₃₀-alkyl)₂, OCH₃, Cl, NH-(CH₂)₁₋₄SO₃H or NH-(CH₂)₁₋₄OH.

Suitable hydroxycoumarins include 6,7-dihydroxycoumarin available from Wako Chemicals, 4-methyl-7-hydroxycoumarin available from Wako Chemicals, 4-methyl-6,7-dihydroxycoumarin available from Wako Chemicals, esculin available from Wako Chemicals, and umbelliferone (4-hydroxycoumarin) available from Wako Chemicals.

15 Aminocoumarins

Aminocoumarins are a class of compounds having the base coumarin structure and having at least one amino moiety.

Aminocoumarins useful in the present inventions include those having formulae (8):

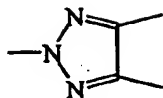


wherein R²⁰⁷ is H, Cl, CH₃, CH₂COOH, CH₂SO₃H, CH₂OSO₃H, or CH₂OPO(OH)OH, R²⁰⁸ is H, phenyl, or COOC₁-C₃₀ alkyl, and R²⁰⁹ and R²¹⁰ are independently H, NH₂, N(C₁-C₃₀alkyl)₂, NHC₁-C₃₀alkyl, or NHCOC₁-C₃₀alkyl.

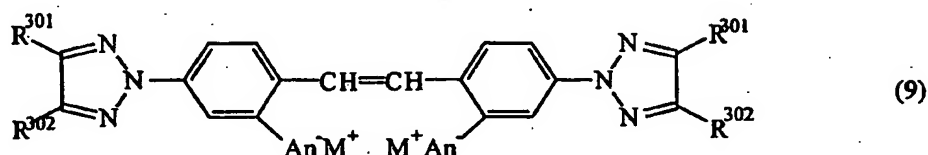
Suitable aminocoumarins include 4-methyl-7,7'-diethylamino coumarin with tradename Calcofluor-RWP available from BASF, 4-methyl-7,7'-dimethylamino coumarin with tradename Calcofluor-LD available from BASF.

30 Triazoles

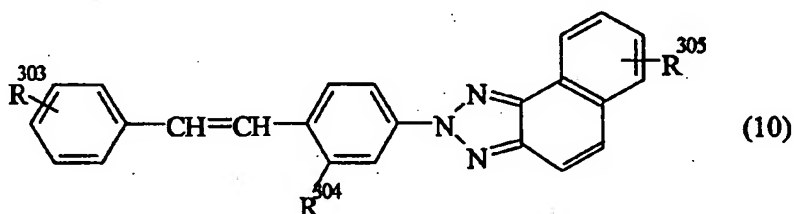
Triazoles are a class of compounds having the following base structure:



Triazoles useful in the present inventions include those having formulae
 5 (9) through (12) and (15) through (20):

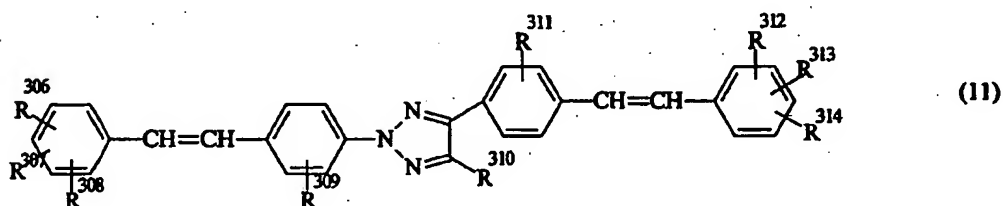


wherein R³⁰¹ and R³⁰², independently, are H, C₁-C₃₀alkyl, phenyl or
 10 monosulfonated phenyl; An⁻ and M are as previously defined, wherein the
 compound has a trans-coplanar orientation or cis-coplanar orientation; preferably
 R³⁰¹ is phenyl, R³⁰² is H and M is sodium; wherein the compound has a trans-
 coplanar orientation;



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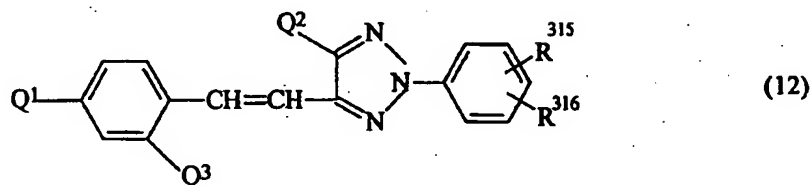
wherein R³⁰³ is H or Cl; R³⁰⁴ is SO₃M, SO₂N(C₁-C₃₀-alkyl)₂, SO₂O-phenyl or
 CN; R³⁰⁵ is H, SO₃M, COOM, OSO₃M, or OPO(OH)OM; and M is as previously
 20 defined, wherein the compound has a trans-coplanar orientation or cis-coplanar
 orientation; preferably R³⁰³ and R³⁰⁵ are H and R³⁰⁴ is SO₃M in which M is
 Na; wherein the compound has a trans-coplanar orientation;



(11)

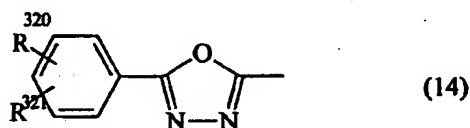
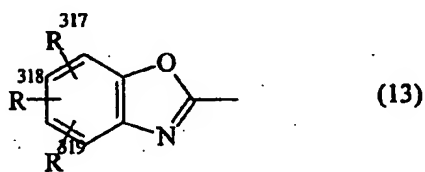
wherein each of R³⁰⁶ and R³¹² independently represents H, a sulfonic acid group or the salts, esters or amides thereof, a carboxylic acid group or the salts, esters or amides thereof, a cyano group, a halogen atom, an unsubstituted or substituted alkylsulfonyl, arylsulfonyl, alkyl, alkoxy, aralkyl, aryl, aryloxy, aralkoxy or cycloalkyl radical, an unsubstituted or substituted 5-membered heterocyclic ring containing 2 to 3 nitrogen atoms or one oxygen atom and 1 or 2 nitrogen atoms, or together with R³⁰⁷ and R³¹³ they represent a methylenedioxy, ethylenedioxy, methylenoxymethylenoxy, trimethylene, tetramethylene, propenylene, butenylene or butadienylene radical, each of R³⁰⁷ and R³¹³ independently represents H, a sulfonic acid group or the salts, esters or amides thereof, a carboxylic acid group or the salts, esters or amides thereof, a cyano group, a halogen atom, an unsubstituted or substituted alkyl or alkoxy radical, or together with R³⁰⁶ and R³¹² represent a methylenedioxy, ethylenedioxy, methylenoxymethylenoxy, trimethylene, tetramethylene, propenylene, butenylene or butadienylene radical, each of R³⁰⁸ and R³¹⁴ independently represents H, a halogen atom or an unsubstituted or substituted alkyl radical, each of R³⁰⁹ and R³¹¹ independently represents H, a halogen atom, a cyano group a sulfonic acid group or the salts, esters or amides thereof, or a carboxylic acid group or the salts, esters or amides thereof, and R³¹⁰ independently represents H, a halogen atom, a cyano group a sulfonic acid group or the salts, alkyl radicals preferably by hydroxy, alkoxy of 1 to 30 carbon atoms, cyano, halogen, carboxy, sulfonic acid groups, carbalkoxy having 1 to 30 carbon atoms in the alkoxy moiety, phenyl or phenoxy; alkoxy radicals can be substituted by hydroxy, alkoxy of 1 to 30 carbon atoms, cyano, halogen, carboxy, carbalkoxy having 1 to 30 carbon atoms in the alkoxy moiety, phenyl or phenoxy; phenyl, phenylalkyl or phenoxy radicals can be substituted by halogen, cyano, carboxy, carbalkoxy having 1 to 30 carbon atoms in the alkoxy moiety, sulfo, or alkyl or alkoxy each of 1 to 30 carbon atoms; wherein the compound has a trans-coplanar orientation or cis-coplanar orientation; possible cycloalkyl radicals are preferably cyclohexyl and cyclopentyl radicals which can be substituted by alkyl of 1 to 30 carbon atoms; possible 5-membered heterocyclic rings are v-triazole, oxazole or 1, 3, 4- oxdiazole radicals which can contain as substituents alkyl radicals of 1 to 4 carbon atoms, halogen, phenyl, carboxy, carbalkoxy having 1 to 30 carbon atoms in the alkoxy moiety, cyano, benzyl, alkoxy of 1 to 30 carbon atoms, phenoxy or sulfo, whilst two adjacent substituents of the triazole and oxazole radicals together are able to

form a substituted or unsubstituted fused benzene nucleus; wherein the compound has a trans-coplanar orientation;



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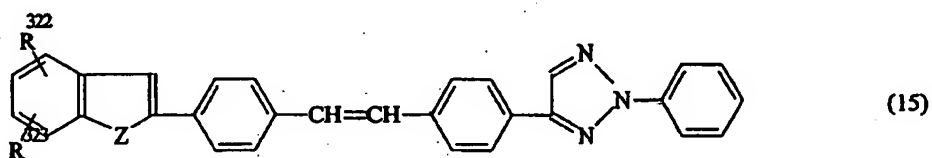
wherein Q¹ denotes one of the ring systems (13) or (14);

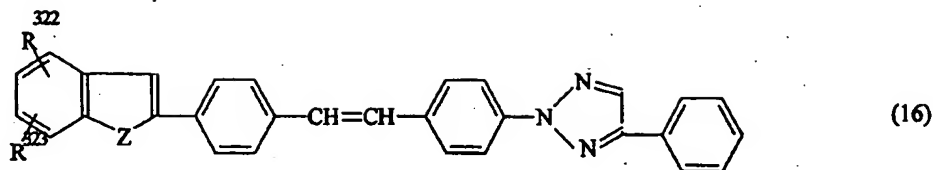


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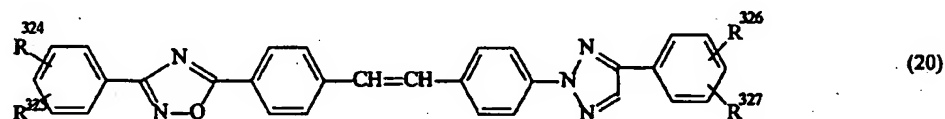
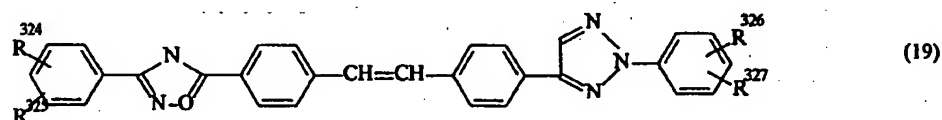
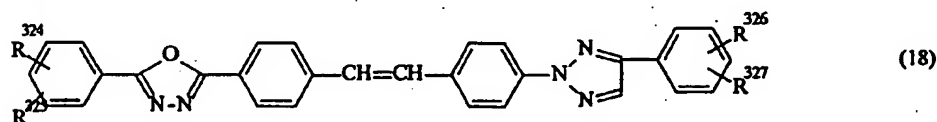
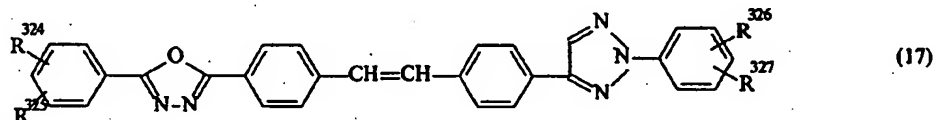
and wherein R³¹⁷ denotes H, alkyl with 1 to 30 carbon atoms, cyclohexyl, phenylalkyl with C₁-C₃₀ carbon atoms in the alkyl part, phenyl, alkoxy with 1 to 30 carbon atoms, or Cl, or, conjointly with R³¹⁸, denotes alkylene with 3 to 30 carbon atoms, R³¹⁸ denotes H or alkyl with 1 to 30 carbon atoms or, conjointly with R³¹⁷, denotes alkylene with 3 to 30 carbon atoms, R³¹⁹ denotes H or methyl, R³²⁰ denotes H, alkyl with 1 to 30 carbon atoms, phenyl, alkoxy with 1 to 30 carbon atoms, or Cl, or, conjointly with R³²¹, denotes a fused benzene ring, R³²¹ denotes H or Cl or conjointly with R³²⁰, denotes a fused benzene ring, R³¹⁵ denotes H, alkyl with 1 to 30 carbon atoms, alkoxy with 1 to 30 carbon atoms or Cl, R³¹⁶ denotes H or Cl, Q² denotes H, Cl alkyl with 1 to 30 carbon atoms or phenyl and Q³ denotes H or Cl; wherein the compound has a trans-coplanar orientation or cis-coplanar orientation, preferably a trans-coplanar orientation;

25





wherein R³²² denotes H, Cl, methyl, phenyl, benzyl, cyclohexyl or methoxy, R³²³ denotes H or methyl and Z denotes O or S; wherein the compound has a trans-coplanar orientation or cis-coplanar orientation, preferably a trans-coplanar orientation; and

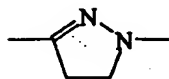


wherein R³²⁴ denotes H, Cl, alkyl with 1 to 30 carbon atoms, phenylalkyl with 1 to 30 carbon atoms, phenyl or alkoxy with 1 to 30 carbon atoms, or R³²⁴ conjointly with R³²⁵ denotes a fused benzene radical, R³²⁵ denotes H or methyl or R³²⁵ conjointly with R³²⁴ denotes a fused benzene radical, R³²⁶ denotes H, alkyl with 1 to 30 carbon atoms, alkoxy with 1 to 30 carbon atoms, Cl, carbalkoxy with 1 to 30 carbon atoms or alkylsulfonyl with 1 to 30 carbon atoms and R³²⁷ denotes H, Cl, methyl or methoxy; wherein the compound has a trans-coplanar orientation or cis-coplanar orientation, preferably a trans-coplanar orientation.

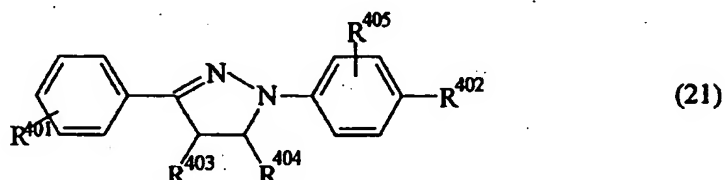
Suitable triazoles include 2-(4-styryl-3-sulfophenyl)-2H-naphtho[1,2-d]triazole (C.I. Fluorescent Brightener 46) with tradename Tinopal RBS available from Ciba Specialty Chemicals.

Pyrazolines

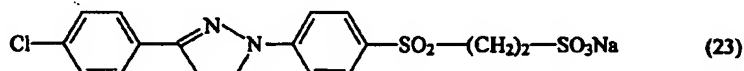
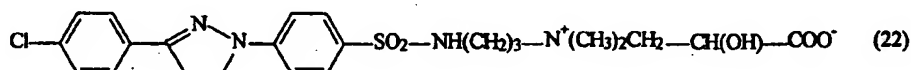
Pyrazolines are a class of compounds having the following base structure:



- 5 Pyrazolines useful in the present invention include those having formulae (21) through (23):



- 10 wherein R⁴⁰¹ is H, Cl or N(C₁-C₃₀-alkyl)₂, R⁴⁰² is H, Cl, SO₃M, SO₂NH₂, SO₂NH-(C₁-C₃₀alkyl), COO-C₁-C₃₀alkyl, SO₂-C₁-C₃₀alkyl, SO₂NH(CH₂)₁₋₄N⁺(CH₃)₃ or SO₂(CH₂)₁₋₄N⁺H(C₁-C₃₀-alkyl)₂An⁻, R⁴⁰³ and R⁴⁰⁴ are the same or different and each is H, C₁-C₃₀alkyl or phenyl and R⁴⁰⁵ is H or Cl; and An⁻ and M are as previously defined, preferably R⁴⁰¹ is Cl, R⁴⁰² is SO₂CH₂
- 15 CH₂N⁺H(C₁-C₄-alkyl)₂An⁻ in which An⁻ is phosphite and R⁴⁰³, R⁴⁰⁴ and R⁴⁰⁵ are each H; and formulae (22) and (23) shown below.



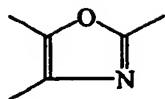
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- Suitable pyrazolines include 1-(4-amidosulfonylphenyl)-3-(4-chlorophenyl)-2-pyrazoline (C.I. Fluorescent Brightener 121) with tradename Blankophor DCB available from Bayer, 1-[4-(2-sulfoethylsulfonyl)phenyl]-3-(4-chlorophenyl)-2-pyrazoline, 1-[4-(2-sulfoethylsulfonyl)phenyl]-3-(3,4-dichloro-6-methylphenyl)-2-pyrazoline, 1-<4-[N-[3-(N,N,N-trimethylammonio)propyl]-amidosulfonyl]phenyl>-3-(4-chlorophenyl)-2-pyrazoline methylsulfate, and 1-<4-[2-[1-methyl-2-(N,N-dimethylamino)ethoxy]ethylsulfonyl]phenyl>-3-(4-chlorophenyl)-2-pyrazoline methylsulfate.

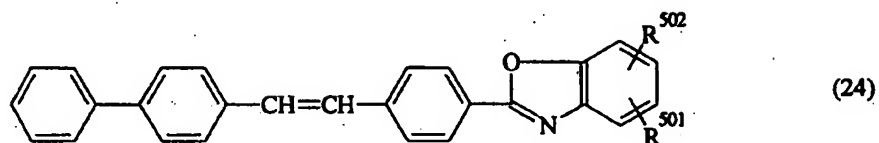
Oxazoles

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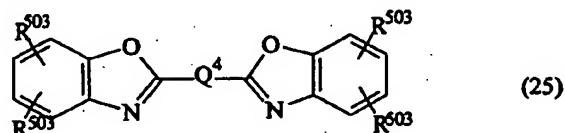
Oxazoles are a class of compounds having the following base structure:



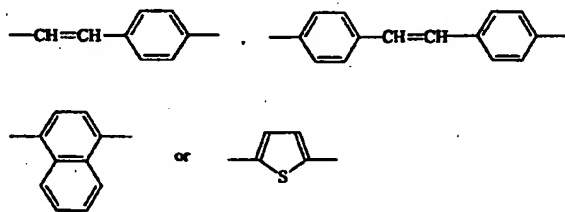
Oxazoles useful in the present inventions include those having formulae
 5 (24), (25), (26) and (27):



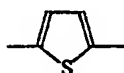
wherein R⁵⁰¹ and R⁵⁰², independently, are H, Cl, C₁-C₃₀alkyl or SO₂-C₁-C₃₀-
 10 alkyl, wherein the compound has a trans-coplanar orientation or cis-coplanar
 orientation; preferably R⁵⁰¹ is 4-CH₃ and R⁵⁰² is 2-CH₃ wherein the compound
 has a trans-coplanar orientation;



15 wherein R⁵⁰³, independently, is H, C(CH₃)₃, C(CH₃)₂-phenyl, C₁-C₃₀alkyl or
 COO-C₁-C₃₀alkyl, preferably H and Q⁴ is -CH=CH-;

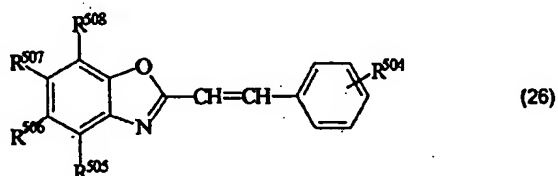


20 preferably

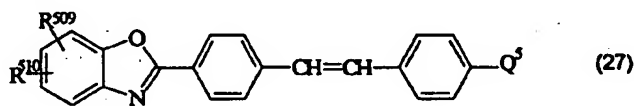


or one group R⁵⁰³ in each ring is 2-methyl and the other R⁵⁰³ is H and Q⁴ is
 25 -CH=CH-; or one group R⁵⁰³ in each ring is 2-C(CH₃)₃ and the other R⁵⁰³ is H;

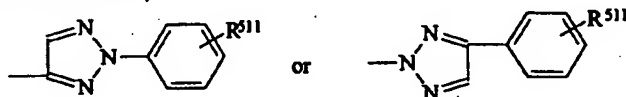
wherein the compound has a trans-coplanar orientation or cis-coplanar orientation, preferably a trans-coplanar orientation;



- 5 wherein R⁵⁰⁴ is CN, Cl, COO-C₁-C₃₀alkyl or phenyl; R⁵⁰⁵ and R⁵⁰⁶ are the atoms required to form a fused benzene ring or R⁵⁰⁶ and R⁵⁰⁸, independently, are H or C₁-C₃₀alkyl; and R⁵⁰⁷ is H, C₁-C₃₀alkyl or phenyl; wherein the compound has a trans-coplanar orientation or cis-coplanar orientation; preferably
- 10 R⁵⁰⁴ is a 4-phenyl group and each of R⁵⁰⁵ to R⁵⁰⁸ is H; wherein the compound has a trans-coplanar orientation; and



- 15 wherein R⁵⁰⁹ denotes H, Cl, alkyl with 1 to 30 carbon atoms, cyclohexyl, phenylalkyl with 1 to 3 carbon atoms in the alkyl part, phenyl or alkoxy with 1 to 30 carbon atoms, R⁵¹⁰ denotes H or alkyl with 1 to 30 carbon atoms, and Q⁵ denotes a radical;

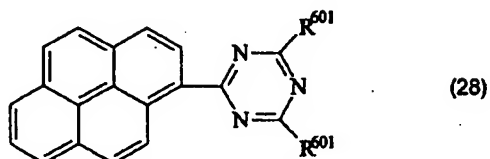


- 20 wherein R⁵¹¹ represents H, alkyl with 1 to 30 carbon atoms, alkoxy with 1 to 30 carbon atoms, Cl, carbalkoxy with 1 to 30 carbon atoms, unsubstituted sulfamoyl or sulfamoyl which is monosubstituted or disubstituted by alkyl or hydroxyalkyl with 1 to 30 carbon atoms or represents alkylsulfonyl with 1 to 30 carbon atoms;
- 25 wherein the compound has a trans-coplanar orientation or cis-coplanar orientation, preferably a trans-coplanar orientation.

Suitable oxazoles include 4,4'-bis(5-methylbenzoxazol-2-yl)stilbene, and 2-(4-methoxycarbonylstyryl)benzoxazole.

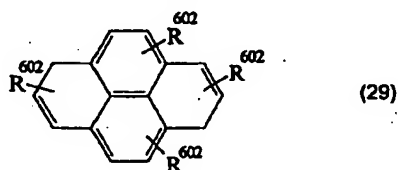
Pyrenes

Pyrenes useful in the present invention include those having formulae (28) and (29):



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wherein each R⁶⁰¹, independently, is C₁-C₃₀alkoxy; preferably methoxy; and



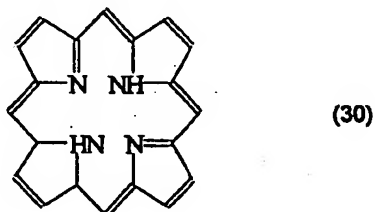
10 wherein each R⁶⁰², independently, is H, OH, or SO₃M, wherein M is as previously defined, sulfonated phenylamino, or anilino.

Suitable pyrenes include 2,4-dimethoxy-6-(1'-pyrenyl)-1,3,5-triazine (C.I. Fluorescent Brightener 179) with tradename Fluolite XMF, 8-hydroxy-1,3,6-pyrenetrisulfonic acid (D&C Green No.8), and 3-hydroxy-5,8,10-trisulphanilic pyrene.

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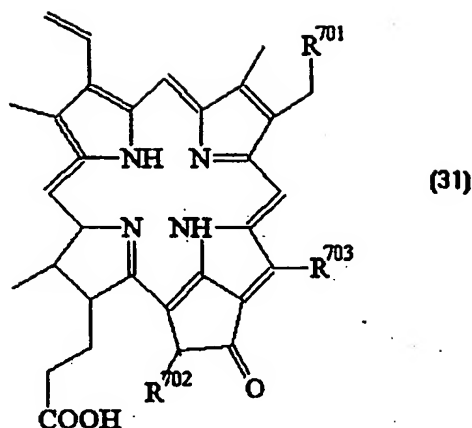
Porphyrins

Porphyrins useful in the present invention include those having formulae (30), (31), and (32):



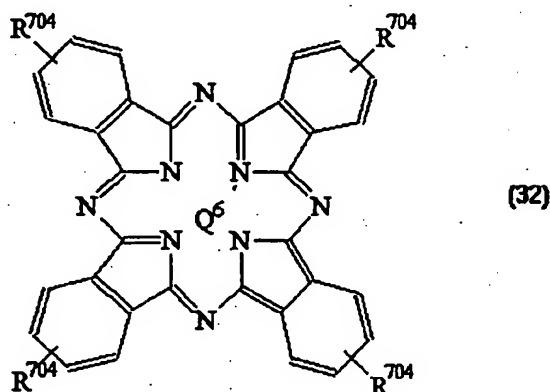
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wherein R^{701} is CH_3 or CHO , R^{702} is H or $\text{COOC}_1\text{-C}_{30}$ alkyl, and R^{703} is H or an alkyl group having 1 to 30 carbons; and

5



wherein each R^{704} , independently, is H , SO_3M , COOM , OSO_3M , or $\text{OPO}(\text{OH})\text{OM}$, wherein M is as previously defined, halide, or alkyl of 1 to 30 carbons; and Q^6 is Cu , Mg , Fe , Cr , Co , or mixtures thereof with cationic charges.

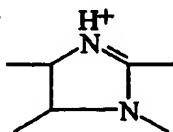
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Suitable porphyrins include porphyrin available from Wako Chemicals and Copper II phthalocyanine available from Wako Chemicals.

Imidazoles

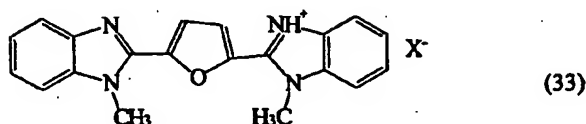
Imidazoles are a class of compounds having the following base structure:

15



Imidazoles useful in the present invention include those having formulae (33):

5



wherein X⁻ is as previously defined.

Suitable imidazoles include those with tradename of C.I. Fluorescence Brightener 352, or Uvtext AT available from Ciba Speciality Chemical.

The optical brightener herein is comprised in an amount which alters the color of the hair to which the composition is applied. Color may be described in terms of three independent variables, i.e., lightness, hue, and saturation. In the present invention, the degree of color alteration is measured by units of color parameters using the colorimetric data calculated from the spectral data measured by either a Macbeth 1500 Reflective Color Meter (supplied by Kollmorgen Corporation) or on a MiniScan XE Spectrophotometer (supplied by Hunter Associates Laboratory). The color parameters are suitably described in the "L,a,b, Color coordinate" as shown in Figure 1. The "L,a,b, Color coordinate" describes the color of any material on the basis of three parameters: L, a, and b. Parameter "L" is a measure of lightness, ranging from 0 (total black) to 100 (total white). The parameters "a" and "b" indicate hue and saturation by the degree of redness (positive a), greenness (negative a) and the degree of yellowness (positive b), blueness (negative b). Parameters "a" and "b" are both zero for neutral colors such as white, gray, and black; i.e.; colors which do not have a hue. The higher the absolute values of a and b, the more saturated or chromatic is the color. The range of a-value varies from -50 which indicates a pure green color, to +50 which indicates a pure red color. The range of b-value varies from -50 which indicates a pure blue color, to +50 which indicates a pure yellow color.

The optical brightener is comprised in an amount which alters the color of the hair to which the composition is applied by from about 0.2 units to about 15 units of a color parameter selected from the group consisting of the a-value, the b-value, and mixtures thereof. When the degree of color alteration is between about 0.2 units to about 15 units of either the a-value, b-value, or both, one perceives that the color of the hair has been altered in such a way that the

unfavorable appearances of damaged hair is alleviated, or that the hair color has become more appealing. For noticeable, yet favorable changes, the optical brightener is comprised in an amount which alters the color of the hair to which the composition is applied by from about 0.5 units to about 10 units, of either or both of the a-value and the b-value.

The color of the hair can be altered towards a more achromatic color, or towards a more saturated or chromatic color. When the color is altered toward decreasing the absolute value of the a-value and/or b-value, the color of the hair is blackened. Such alteration may be effective for consumers who consider perfect black hair as appealing and/or undamaged. On the other hand, when the color is altered toward increasing the absolute value of either or both of the a-value or b-value, the hair is altered to a more saturated or chromatic color. Such alteration may be effective for consumers who consider a saturated or chromatic color as appealing. The kind and amount of optical brighteners are selected according to their absorption/emission spectrums to provide the targeted degree and type of color alteration. Two or more optical brighteners may be included in the composition to provide the targeted degree and type of color alteration. It is known to one skilled in the art that such color alteration of the hair can also be achieved by applying dye-stuff to the hair. In the present invention, the targeted degree and type of color alteration is achieved without the aid of any dye-stuff. Optionally, dye-stuff may be included in the composition of the present invention in an amount which would not cause damage or dullness to the hair. Dye-stuff, however, is not an essential component.

Any color of hair can be described by the L,a,b color parameters. The composition of the present invention can alter the color of hair having any color parameter. Preferably, the alteration of color occurs within 20 units from the zero point of the a-value axis and b-value axis. The alteration of color is particularly noticeable when the alteration occurs within 10 units, preferably 5 units from the zero point of the a-value axis and b-value axis.

The specific amount of optical brightener to be included in the hair care composition varies depending on the characteristics of the optical brightener, the product form in which the optical brightener is included, and other components included in the composition. For example, optical brighteners which have a relatively strong emission will alter the color parameters with less amount than those which have a relatively weak emission. In another example, product forms aimed to be left on the hair may be formulated with less amount of optical

brightener than those product forms aimed to be rinsed off from the hair. In another example, compositions including components which provide a strong deposition to the hair may have less amount of optical brightener than compositions which do not include such components. The composition is preferably formulated so that the optical brightener applied to the hair is from about 1ppm to about 2%, preferably 10ppm to about 1%, more preferably from about 100ppm to about 0.3% by weight of the hair. Preferably, optical brighteners herein are included in the hair care composition of the present invention at a level by weight of from about 0.001% to about 20%, more preferably from about 0.01% to about 10%.

The hair care compositions of the present invention herein including optical brighteners provide benefits to the hair in three areas. First, optical brighteners herein alter the color of the hair by emitting light in the visible range, as explained in detail above. Second, optical brighteners herein enhance the shine of the hair by emitting light in the visible range. Third, optical brighteners herein protect the hair from ultraviolet light by absorbing ultraviolet light.

The hair care compositions of the present invention can be used daily. By repetitive daily use of the composition of the present invention, the effect of hair color alteration is enhanced. On the other hand, the color of the hair altered by the composition of the present invention may be returned to its original color by cleansing the hair with regular shampoo one to about several times.

HAIR CARE ACTIVE

The hair care compositions of the present invention comprise a component which characterizes the form and function of the product. Product forms useful herein include, but are not limited to, shampoo, conditioner, treatment, mousse, spray, lotion, gel, and cream products, all of which can be designed for rinse-off or leave-on convenience. Product functions useful herein include, but are not limited to, cleaning, conditioning, and styling products. To make these products, a hair care active selected from the group consisting of a deterative surfactant, a conditioning agent, a fixative polymer, and mixtures thereof are comprised. The hair care actives and their levels are selected by one skilled in the art depending on the desired characteristic of the product.

DETERSIVE SURFACTANT

Compositions herein may include a deterative surfactant. The deterative surfactants herein are those suitable for cleansing the hair. Deterative surfactants

useful herein include anionic surfactants, amphoteric and zwitterionic surfactants, and nonionic surfactants.

The deterative surfactants, when present, are preferably included at a level of from about 0.01% to about 75% by weight of the composition. Two or more surfactants can be used.

Anionic Surfactant

Anionic surfactants useful herein include alkyl and alkyl ether sulfates. These materials have the respective formulae ROSO_3M and $\text{RO}(\text{C}_2\text{H}_4\text{O})_x\text{SO}_3\text{M}$, wherein R is alkyl or alkenyl of from about 8 to about 30 carbon atoms, x is 1 to about 10, and M is hydrogen or a cation such as ammonium, alkanolammonium (e.g., triethanolammonium), a monovalent metal cation (e.g., sodium and potassium), or a polyvalent metal cation (e.g., magnesium and calcium). Preferably, M should be chosen such that the anionic surfactant component is water soluble. The anionic surfactant or surfactants should be chosen such that the Krafft temperature is about 15°C or less, preferably about 10°C or less, and more preferably about 0°C or less. It is also preferred that the anionic surfactant be soluble in the composition hereof.

Krafft temperature refers to the point at which solubility of an ionic surfactant becomes determined by crystal lattice energy and heat of hydration, and corresponds to a point at which solubility undergoes a sharp, discontinuous increase with increasing temperature. Each type of surfactant will have its own characteristic Krafft temperature. Krafft temperature for ionic surfactants is, in general, well known and understood in the art. See, for example, Myers, Drew, Surfactant Science and Technology, pp. 82-85, VCH Publishers, Inc. (New York, New York, USA), 1988 (ISBN 0-89573-399-0), which is incorporated by reference herein in its entirety.

In the alkyl and alkyl ether sulfates described above, preferably R has from about 8 to about 18 carbon atoms in both the alkyl and alkyl ether sulfates. The alkyl ether sulfates are typically made as condensation products of ethylene oxide and monohydric alcohols having from about 8 to about 24 carbon atoms. The alcohols can be derived from fats, e.g., coconut oil, palm oil, tallow, or the like, or the alcohols can be synthetic. Lauryl alcohol and straight chain alcohols derived from coconut oil and palm oil are preferred herein. Such alcohols are reacted with 1 to about 10, and especially about 3, molar proportions of ethylene oxide and the resulting mixture of molecular species having, for example, an

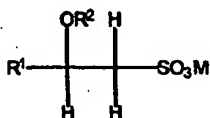
average of 3 moles of ethylene oxide per mole of alcohol, is sulfated and neutralized.

Specific examples of alkyl ether sulfates which can be used are sodium and ammonium salts of coconut alkyl triethylene glycol ether sulfate; tallow alkyl triethylene glycol ether sulfate, and tallow alkyl hexaoxyethylene sulfate. Highly preferred alkyl ether sulfates are those comprising a mixture of individual compounds, said mixture having an average alkyl chain length of from about 8 to about 16 carbon atoms and an average degree of ethoxylation of from 1 to about 4 moles of ethylene oxide. Such a mixture also comprises from 0% to about 20% by weight C_{12-13} compounds; from about 60% to about 100% by weight of $C_{14-15-16}$ compounds, from 0% to about 20% by weight of $C_{17-18-19}$ compounds; from about 3% to about 30% by weight of compounds having a degree of ethoxylation of 0; from about 45% to about 90% by weight of compounds having a degree of ethoxylation of from 1 to about 4; from about 10% to about 25% by weight of compounds having a degree of ethoxylation of from about 4 to about 8; and from about 0.1% to about 15% by weight of compounds having a degree of ethoxylation greater than about 8.

Other suitable anionic surfactants are the water-soluble salts of organic, sulfuric acid reaction products of the general formula $[R^1-SO_3-M]$ where R^1 is selected from the group consisting of a straight or branched chain, saturated aliphatic hydrocarbon radical having from about 8 to about 24, preferably about 8 to about 18, carbon atoms; and M is as previously described above in this section. Examples of such surfactants are the salts of an organic sulfuric acid reaction product of a hydrocarbon of the methane series, including iso-, neo-, and n-paraffins, having about 8 to about 24 carbon atoms, preferably about 8 to about 18 carbon atoms and a sulfonating agent, e.g., SO_3 , H_2SO_4 , obtained according to known sulfonation methods, including bleaching and hydrolysis. Preferred are alkali metal and ammonium sulfonated C_{8-18} n-paraffins.

Still other suitable anionic surfactants are the reaction products of fatty acids esterified with isethionic acid and neutralized with sodium hydroxide where, for example, the fatty acids are derived from coconut or palm oil; or sodium or potassium salts of fatty acid amides of methyl tauride in which the fatty acids, for example, are derived from coconut oil. Other similar anionic surfactants are described in U.S. Patents 2,486,921, 2,486,922, and 2,396,278, which are incorporated by reference herein in their entirety.

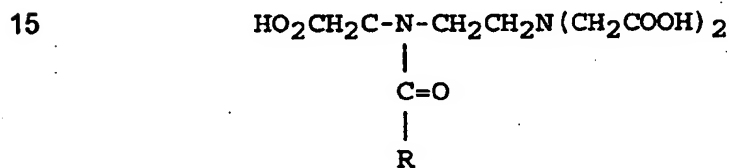
Another class of anionic surfactants suitable for use in the shampoo compositions are the β -alkyloxy alkane sulfonates. These compounds have the following formula:



- 5 where R¹ is a straight chain alkyl group having from about 6 to about 20 carbon atoms, R² is a lower alkyl group having from about 1, preferred, to about 3 carbon atoms, and M is as hereinbefore described. Many other anionic surfactants suitable for use in the shampoo compositions are described in
- 10 McCutcheon's, Emulsifiers and Detergents, 1989 Annual, published by M. C. Publishing Co., and in U.S. Patent 3,929,678, which descriptions are incorporated herein by reference in their entirety. Preferred anionic surfactants for use in the shampoo compositions include ammonium lauryl sulfate, ammonium laureth sulfate, triethylamine lauryl sulfate, triethylamine laureth sulfate,
- 15 sulfate, triethanolamine lauryl sulfate, triethanolamine laureth sulfate, monoethanolamine lauryl sulfate, monoethanolamine laureth sulfate, diethanolamine lauryl sulfate, diethanolamine laureth sulfate, lauric monoglyceride sodium sulfate, sodium lauryl sulfate, sodium laureth sulfate, potassium lauryl sulfate, potassium laureth sulfate, ammonium cocoyl sulfate,
- 20 ammonium lauroyl sulfate, sodium cocoyl sulfate, sodium lauroyl sulfate, potassium cocoyl sulfate, potassium lauryl sulfate, triethanolamine lauryl sulfate, triethanolamine lauryl sulfate, monoethanolamine cocoyl sulfate, monoethanolamine lauryl sulfate, sodium tridecyl benzene sulfonate, and sodium dodecyl benzene sulfonate, and mixtures thereof.
- 25 Other anionic surfactants for use herein include polyhydrophilic anionic surfactants. By "polyhydrophilic" herein, is meant a surfactant that has at least two hydrophilic groups which provide a hydrophilic nature. Polyhydrophilic surfactants useful herein are only those having at least two hydrophilic groups in the molecule, and is not intended to encompass those which only have one
- 30 hydrophilic group. One molecule of the polyhydrophilic anionic surfactant herein may comprise the same hydrophilic groups, or different hydrophilic groups. Specifically, the polyhydrophilic anionic surfactants comprise at least one group selected from the group consisting of carboxy, hydroxy, sulfate, sulfonate, and phosphate. Suitable polyhydrophilic anionic surfactants are those which

comprise at least one of a carboxy, sulfate, or sulfonate group, more preferably those which comprise at least one carboxy group.

Nonlimiting examples of polyhydrophilic anionic surfactants include N-acyl-L-glutamates such as N-cocoyl-L-glutamate and, N-lauroyl-L-glutamate, laurimino dipropionate, N-acyl-L-aspartate, di-(N-lauroyl N-methyl taurate), polyoxyethylene laurylsulfosuccinate, disodium N-octadecylsulfosuccinate; disodium lauryl sulfosuccinate; diammonium lauryl sulfosuccinate; tetra sodium N-(1,2-dicarboxyethyl)-N-octadecylsulfosuccinate; the diamyl ester of sodium sulfosuccinic acid; the dihexyl ester of sodium sulfosuccinic acid; and the dioctyl ester of sodium sulfosuccinic acid, and 2-cocoalkyl N-carboxyethyl N-carboxyethoxyethyl imidazolinium betaine, lauroamphohydroxypropylsulfonate, cocoglyceryl ether salts, cocoglyceride sulfate, lauroyl isethionate, lauroamphoacetate, and those of the following formula:



wherein R is an alkyl of 8 to 18 carbons. Other polyhydrophilic anionic surfactants include olefin sulfonates having about 10 to about 24 carbon atoms. The term "olefin sulfonates" is used herein to mean compounds which can be produced by the sulfonation of alpha-olefins by means of uncomplexed sulfur trioxide, followed by neutralization of the acid reaction mixture in conditions such that any sulfones which have been formed in the reaction are hydrolyzed to give the corresponding hydroxy-alkanesulfonates. The sulfur trioxide can be liquid or gaseous, and is usually, but not necessarily, diluted by inert diluents, for example by liquid SO₂, chlorinated hydrocarbons, etc., when used in the liquid form, or by air, nitrogen, gaseous SO₂, etc., when used in the gaseous form. The α-olefins from which the olefin sulfonates are derived are mono-olefins having about 8 to about 24 carbon atoms, preferably about 10 to about 16 carbon atoms. Preferably, they are straight chain olefins. In addition to the true alkene sulfonates and a proportion of hydroxy-alkanesulfonates, the olefin sulfonates can contain minor amounts of other materials, such as alkene disulfonates depending upon the reaction conditions, proportion of reactants, the nature of the starting olefins and impurities in the olefin stock and side reactions during the sulfonation process. A specific α-olefin sulfonate mixture of the above type is

described more fully in U.S. Patent 3,332,880, to Pflaumer and Kessler, issued July 25, 1967, which is incorporated by reference herein in its entirety.

Another class of polyhydrophilic anionic surfactants are amino acid surfactants which are surfactants that have the basic chemical structure of an amino acid compound, i.e., that contains a structural component of one of the naturally-occurring amino acids. It is understood by the artisan that some surfactants may be regarded as both a polyhydrophilic anionic surfactant, and an amino acid surfactant. These surfactants are suitable anionic surfactants.

Nonlimiting examples of amino acid surfactants include, N-cocoylalaninate, N-acyl-N-methyl- β -alanate, N-acylsarcosinate; N-alkylamino propionates and N-alkyliminodipropionates, specific examples of which include N-lauryl- β -amino propionic acid or salts thereof, and N-lauryl- β -iminodipropionate, N-acyl-DL-alaninate, sodium lauryl sarcosinate, sodium lauroyl sarcosinate, lauryl sarcosine, cocoyl sarcosine, N-acyl-N-methyl taurate, lauroyl taurate, and lauroyl lactylate.

Commercially available anionic surfactants suitable are N-acyl-L-glutamate with a tradename AMISOFT CT-12S, N-acyl potassiumglycine with a tradename AMILITE GCK-12, lauroyl glutamate with a tradename AMISOFT LS-11, and N-acyl-DL-alaninate with tradename AMILITE ACT12 supplied by Ajinomoto; acylaspartate with tradenames ASPARACK and AAS supplied by Mitsubishi Chemical; and acyl derivatives of tradename ED3A supplied by Hampshire Chemical Corp.

Optionally the counter ion of anionic surfactants may be polyvalent cations. It has been found that these anionic surfactants, along with the cationic conditioning agents, and polyvalent metal cations as described later, form a coacervate in the compositions. Cationic conditioning agents may be included in the present composition to provide a shampoo which both cleanse and condition the hair from a single product.

Coacervate formulation is dependent upon a variety of criteria such as molecular weight, component concentration, and ratio of interacting ionic components ionic strength, charge density of the cationic and anionic components, pH, and temperature. Coacervate systems and the effect of these parameters are known in the art.

It is believed to be particularly advantageous, for the anionic surfactants and the polyvalent metal cations at certain levels to be present with the cationic conditioning agents in a coacervate phase. The coacervates formed in the

compositions are believed to readily deposit on the hair upon diluting the coacervate with abundant water, i.e., rinsing of the shampoo.

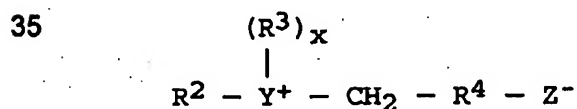
Without being bound by theory, it is believed that the coacervates provide two major effects to the present shampoo composition. First, it reduces the Critical Micelle Concentration (hereinafter "CMC") of the composition. The reduction of the CMC relates to reduction of the surface tension, thereby improving lather performance. Second, the existence of the anionic surfactants along with the polyvalent metal cations expand the coacervate region in the composition. As the cationic conditioning agents in the composition are mainly delivered to the hair via these coacervates, expansion of the coacervate region results in delivery of more cationic conditioning agents to the hair. Consequently, compositions which both cleanse and condition the hair from a single product, which have improved overall conditioning benefits and improved lathering are provided.

Techniques of analysis of formation of complex coacervates are known in the art. For example, microscopic analysis of the shampoo compositions, at any chosen stage of dilution, can be utilized to identify whether a coacervate phase has formed. Such coacervate phase will be identifiable as an additional emulsified phase in the composition. The use of dyes can aid in distinguishing the coacervate phase from other insoluble phases dispersed in the shampoo composition.

Amphoteric And Zwitterionic Surfactant

Amphoteric surfactants for use herein include the derivatives of aliphatic secondary and tertiary amines in which the aliphatic radical is straight or branched and one of the aliphatic substituents contains from about 8 to about 18 carbon atoms and one contains an anionic water solubilizing group, e.g., carboxy, sulfonate, sulfate, phosphate, or phosphonate.

Zwitterionic surfactants for use herein include the derivatives of aliphatic quaternary ammonium, phosphonium, and sulfonium compounds, in which the aliphatic radicals are straight or branched, and wherein one of the aliphatic substituents contains from about 8 to about 18 carbon atoms and one contains an anionic group, e.g., carboxy, sulfonate, sulfate, phosphate, or phosphonate. A general formula for these compounds is:

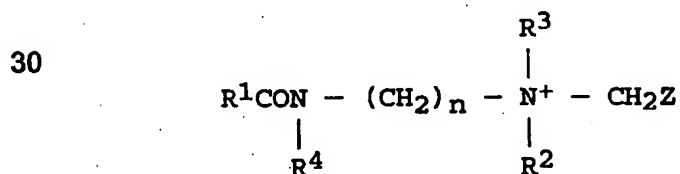


where R^2 contains an alkyl, alkenyl, or hydroxy alkyl radical of from about 8 to about 18 carbon atoms, from 0 to about 10 ethylene oxide moieties and from 0 to about 1 glyceryl moiety; Y is selected from the group consisting of nitrogen, phosphorus, and sulfur atoms; R^3 is an alkyl or monohydroxyalkyl group containing 1 to about 3 carbon atoms; X is 1 when Y is a sulfur atom, and 2 when Y is a nitrogen or phosphorus atom; R^4 is an alkylene or hydroxyalkylene of from 1 to about 4 carbon atoms and Z is a radical selected from the group consisting of carboxylate, sulfonate, sulfate, phosphonate, and phosphate groups.

Examples of amphoteric and zwitterionic surfactants also include sultaines and amidosultaines. Sultaines, including amidosultaines, include for example, cocodimethylpropylsultaine, stearyldimethylpropylsultaine, lauryl-bis-(2-hydroxyethyl)propylsultaine and the like; and the amidosultaines such as cocamidodimethylpropylsultaine, stearylamidododimethylpropylsultaine, laurylamido-bis-(2-hydroxyethyl)propylsultaine, and the like. Preferred are amidohydroxysultaines such as the C₈-C₁₈ hydrocarbylamidopropyl hydroxysultaines, especially C₈-C₁₄ hydrocarbylamidopropylhydroxysultaines, e.g., laurylamidopropylhydroxysultaine and cocamidopropylhydroxysultaine. Other sultaines are described in U.S. Patent 3,950,417, which is incorporated herein by reference in its entirety.

Other suitable amphoteric surfactants are the aminoalkanoates of the formula $RNH(CH_2)_nCOOM$, the iminodialkanoates of the formula $RN[(CH_2)_mCOOM]_2$ and mixtures thereof; wherein n and m are numbers from 1 to about 4, R is C₈ - C₂₂ alkyl or alkenyl, and M is hydrogen, alkali metal, alkaline earth metal, ammonium or alkanolammonium.

Other suitable amphoteric surfactants include those represented by the formula :



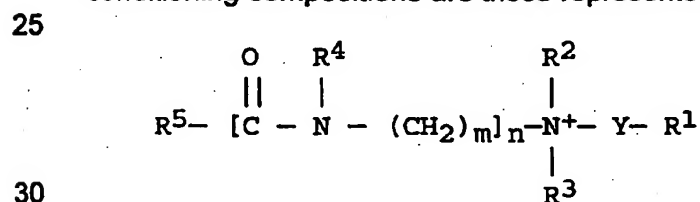
wherein R^1 is C₈ - C₂₂ alkyl or alkenyl, preferably C₈ - C₁₆, R^2 and R^3 is independently selected from the group consisting of hydrogen, $-CH_2CO_2M$, $-CH_2CH_2OH$, $-CH_2CH_2OCH_2CH_2COOM$, or $-(CH_2CH_2O)_mH$ wherein m is an integer from 1 to about 25, and R^4 is hydrogen, $-CH_2CH_2OH$, or

CH₂CH₂OCH₂CH₂COOM, Z is CO₂M or CH₂CO₂M, n is 2 or 3, preferably 2, M is hydrogen or a cation, such as alkali metal (e.g., lithium, sodium, potassium), alkaline earth metal (beryllium, magnesium, calcium, strontium, barium), or ammonium. This type of surfactant is sometimes classified as an imidazoline-type amphoteric surfactant, although it should be recognized that it does not necessarily have to be derived, directly or indirectly, through an imidazoline intermediate. Suitable materials of this type are marketed under the tradename MIRANOL and are understood to comprise a complex mixture of species, and can exist in protonated and non-protonated species depending upon pH with respect to species that can have a hydrogen at R². All such variations and species are meant to be encompassed by the above formula.

Examples of surfactants of the above formula are monocarboxylates and di-carboxylates. Examples of these materials include cocoamphocarboxypropionate, cocoamphocarboxypropionic acid, cocoamphocarboxyglycinate (alternately referred to as cocoamphodiacetate), and cocoamphoacetate.

Commercial amphoteric surfactants include those sold under the trade names MIRANOL C2M CONC. N.P., MIRANOL C2M CONC. O.P., MIRANOL C2M SF, MIRANOL CM SPECIAL (Miranol, Inc.); ALKATERIC 2CIB (Alkaril Chemicals); AMPHOTERGE W-2 (Lonza, Inc.); MONATERIC CDX-38, MONATERIC CSH-32 (Mona Industries); REWOTERIC AM-2C (Rewo Chemical Group); and SCHERCOTERIC MS-2 (Scher Chemicals).

Betaine surfactants, i.e. zwitterionic surfactants, suitable for use in the conditioning compositions are those represented by the formula:



wherein: R¹ is a member selected from the group consisting of COOM and CH(OH)CH₂SO₃M

R² is lower alkyl or hydroxyalkyl; R³ is lower alkyl or hydroxyalkyl; R⁴ is a member selected from the group consisting of hydrogen and lower alkyl; R⁵ is higher alkyl or alkenyl; Y is lower alkyl, preferably methyl; m is an integer from 2 to 7, preferably from 2 to 3; n is the integer 1 or 0; M is hydrogen or a cation, as previously described, such as an alkali metal, alkaline earth metal, or ammonium.

The term "lower alkyl" or "hydroxyalkyl" means straight or branch chained, saturated, aliphatic hydrocarbon radicals and substituted hydrocarbon radicals having from one to about three carbon atoms such as, for example, methyl, ethyl, propyl, isopropyl, hydroxypropyl, hydroxyethyl, and the like. The term "higher alkyl or alkenyl" means straight or branch chained saturated (i.e., "higher alkyl") and unsaturated (i.e., "higher alkenyl") aliphatic hydrocarbon radicals having from about 8 to about 20 carbon atoms such as, for example, lauryl, cetyl, stearyl, oleyl, and the like. It should be understood that the term "higher alkyl or alkenyl" includes mixtures of radicals which may contain one or more intermediate linkages such as ether or polyether linkages or non-functional substituents such as hydroxyl or halogen radicals wherein the radical remains of hydrophobic character.

Examples of surfactant betaines of the above formula wherein n is zero which are useful herein include the alkylbetaines such as cocodimethylcarboxymethylbetaine, lauryldimethylcarboxymethylbetaine, lauryldimethyl- α -carboxyethylbetaine, cetyldimethylcarboxymethylbetaine, lauryl-bis-(2-hydroxyethyl)-carboxymethylbetaine, stearyl-bis-(2-hydroxypropyl)carboxymethylbetaine, oleyldimethyl- γ -carboxypropylbetaine, lauryl-bis-(2-hydroxypropyl)- α -carboxyethylbetaine, etc. The sulfobetaines may be represented by cocodimethylsulfopropylbetaine, stearyldimethylsulfopropylbetaine, lauryl-bis-(2-hydroxyethyl)-sulfopropylbetaine, and the like.

Specific examples of amido betaines and amidosulfobetaines useful in the conditioning compositions include the amidocarboxybetaines, such as cocamidodimethylcarboxymethylbetaine, laurylamidodimethylcarboxymethylbetaine, cetylamidodimethylcarboxymethylbetaine, laurylamido-bis-(2-hydroxyethyl)carboxymethylbetaine, cocamido-bis-(2-hydroxyethyl)-carboxymethylbetaine, etc. The amidosulfobetaines may be represented by cocamidodimethylsulfopropylbetaine, stearylamidodimethylsulfopropylbetaine, laurylamido-bis-(2-hydroxyethyl)-sulfopropylbetaine, and the like.

Nonionic Surfactant

The compositions of the present invention can comprise a nonionic surfactant. Nonionic surfactants include those compounds produced by condensation of alkylene oxide groups, hydrophilic in nature, with an organic hydrophobic compound, which may be aliphatic or alkyl aromatic in nature.

Preferred nonlimiting examples of nonionic surfactants for use in the shampoo compositions include the following:

(1) polyethylene oxide condensates of alkyl phenols, e.g., the condensation products of alkyl phenols having an alkyl group containing from about 6 to about 20 carbon atoms in either a straight chain or branched chain configuration, with ethylene oxide, the said ethylene oxide being present in amounts equal to from about 10 to about 60 moles of ethylene oxide per mole of alkyl phenol;

(2) those derived from the condensation of ethylene oxide with the product resulting from the reaction of propylene oxide and ethylene diamine products;

(3) condensation products of aliphatic alcohols having from about 8 to about 18 carbon atoms, in either straight chain or branched chain configurations, with ethylene oxide, e.g., a coconut alcohol ethylene oxide condensate having from about 10 to about 30 moles of ethylene oxide per mole of coconut alcohol, the coconut alcohol fraction having from about 8 to about 14 carbon atoms;

(4) long chain tertiary amine oxides of the formula $[R^1R^2R^3N \rightarrow O]$ where R^1 contains an alkyl, alkenyl or monohydroxy alkyl radical of from about 8 to about 18 carbon atoms, from 0 to about 10 ethylene oxide moieties, and from 0 to about 1 glyceryl moiety, and R^2 and R^3 contain from about 1 to about 3 carbon atoms and from 0 to about 1 hydroxy group, e.g., methyl, ethyl, propyl, hydroxyethyl, or hydroxypropyl radicals;

(5) long chain tertiary phosphine oxides of the formula $[RR'R''P \rightarrow O]$ where R contains an alkyl, alkenyl or monohydroxyalkyl radical ranging from about 8 to about 18 carbon atoms in chain length, from 0 to about 10 ethylene oxide moieties and from 0 to 1 glyceryl moieties and R' and R'' are each alkyl or monohydroxyalkyl groups containing from about 1 to about 3 carbon atoms;

(6) long chain dialkyl sulfoxides containing one short chain alkyl or hydroxy alkyl radical of from 1 to about 3 carbon atoms (usually methyl) and one long hydrophobic chain which include alkyl, alkenyl, hydroxy alkyl, or keto alkyl radicals containing from about 8 to about 20 carbon atoms, from 0 to about 10 ethylene oxide moieties and from 0 to 1 glyceryl moieties;

(7) alkyl polysaccharide (APS) surfactants (e.g. alkyl polyglycosides), examples of which are described in U.S. Patent 4,565,647, which is incorporated herein by reference in its entirety, and which discloses APS surfactants having a hydrophobic group with about 6 to about 30 carbon atoms and a polysaccharide (e.g., polyglycoside) as the hydrophilic group; optionally, there can be a polyalkylene-oxide group joining the hydrophobic and hydrophilic moieties; and the alkyl group (i.e., the hydrophobic moiety) can be saturated or unsaturated,

branched or unbranched, and unsubstituted or substituted (e.g., with hydroxy or cyclic rings); a preferred material is alkyl polyglucoside which is commercially available from Henkel, ICI Americas, and Seppic; and

- (8) polyoxyethylene alkyl ethers such as those of the formula
 5 RO(CH₂CH₂)_nH and polyethylene glycol (PEG) glyceryl fatty esters, such as those of the formula R(O)OCH₂CH(OH)CH₂(OCH₂CH₂)_nOH, wherein n is from 1 to about 200, preferably from about 20 to about 100, and R is an alkyl having from about 8 to about 22 carbon atoms.

CONDITIONING AGENTS

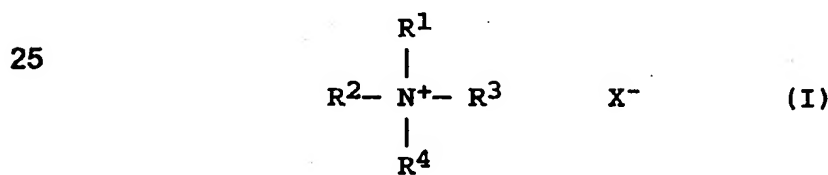
- 10 Compositions herein may include a conditioning agent. Conditioning agents useful herein are those suitable for conditioning or treating the hair. Conditioning agents useful herein include cationic surfactants, high melting point compounds, oily compounds, cationic polymers, silicone compounds, and nonionic polymers.

- 15 The conditioning agents, when present, are preferably included at a level of from about 0.01% to about 30% by weight of the composition. When two or more conditioning agents are used, preferably each conditioning agents does not exceed about 20% by weight of the composition.

Cationic Surfactant

- 20 The cationic surfactants useful herein are any known to the artisan.

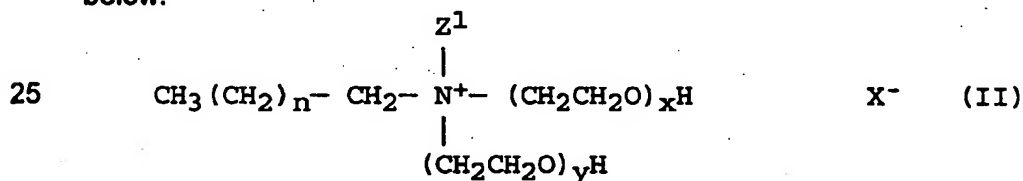
Among the cationic surfactants useful herein are those corresponding to the general formula (I):



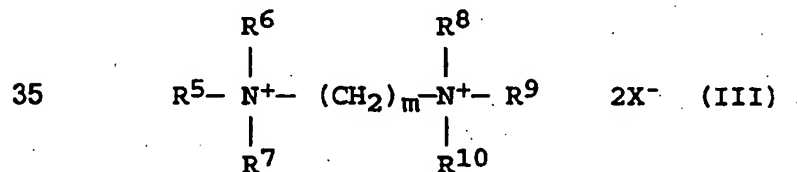
- 30 wherein at least one of R¹, R², R³, and R⁴ is selected from an aliphatic group of from 8 to 30 carbon atoms or an aromatic, alkoxy, polyoxyalkylene, alkylamido, hydroxyalkyl, aryl or alkylaryl group having up to about 22 carbon atoms, the remainder of R¹, R², R³, and R⁴ are independently selected from an aliphatic group of from 1 to about 22 carbon atoms or an aromatic, alkoxy,
 35 polyoxyalkylene, alkylamido, hydroxyalkyl, aryl or alkylaryl group having up to about 22 carbon atoms; and X is a salt-forming anion such as those selected from halogen, (e.g. chloride, bromide), acetate, citrate, lactate, glycolate, phosphate, nitrate, sulfonate, sulfate, alkylsulfate, and alkyl sulfonate radicals.

The aliphatic groups can contain, in addition to carbon and hydrogen atoms, ether linkages, and other groups such as amino groups. The longer chain aliphatic groups, e.g., those of about 12 carbons, or higher, can be saturated or unsaturated. Preferred is when R¹, R², R³, and R⁴ are independently selected from C₁ to about C₂₂ alkyl. Nonlimiting examples of cationic surfactants useful include the materials having the following CTFA designations: quaternium-8, quaternium-24, quaternium-26, quaternium-27, quaternium-30, quaternium-33, quaternium-43, quaternium-52, quaternium-53, quaternium-56, quaternium-60, quaternium-62, quaternium-70, quaternium-72, quaternium-75, quaternium-77, quaternium-78, quaternium-80, quaternium-81, quaternium-82, quaternium-83, quaternium-84, and mixtures thereof.

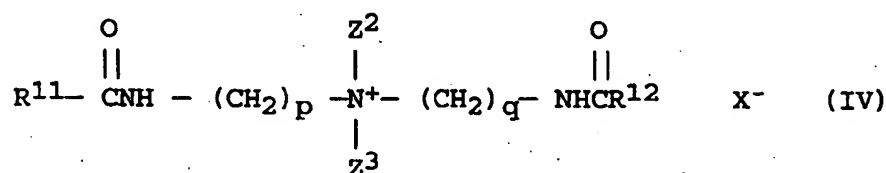
Also preferred are hydrophilically substituted cationic surfactants in which at least one of the substituents contain one or more aromatic, ether, ester, amido, or amino moieties present as substituents or as linkages in the radical chain, wherein at least one of the R¹ - R⁴ radicals contain one or more hydrophilic moieties selected from alkoxy (preferably C₁ - C₃ alkoxy), polyoxyalkylene (preferably C₁ - C₃ polyoxyalkylene), alkylamido, hydroxyalkyl, alkylester, and combinations thereof. Preferably, the hydrophilically substituted cationic conditioning surfactant contains from 2 to about 10 nonionic hydrophile moieties located within the above stated ranges. Preferred hydrophilically substituted cationic surfactants include those of the formula (II) through (VII) below:



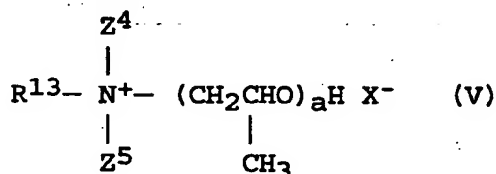
wherein n is from 8 to about 28, x+y is from 2 to about 40, Z¹ is a short chain alkyl, preferably a C₁ - C₃ alkyl, more preferably methyl, or - (CH₂CH₂O)_zH wherein x+y+z is up to 60, and X is a salt forming anion as defined above;



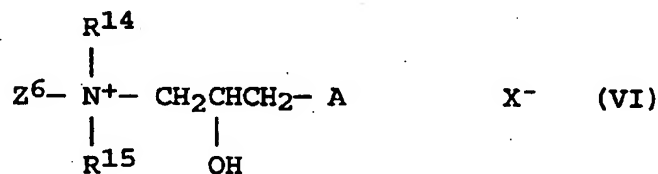
wherein m is 1 to 5, one or more of R⁵, R⁶, and R⁷ are independently an C₁ - C₃₀ alkyl, the remainder are -CH₂CH₂OH, one or two of R⁸, R⁹, and R¹⁰ are independently an C₁ - C₃₀ alkyl, and remainder are -CH₂CH₂OH, and X is a salt forming anion as mentioned above;



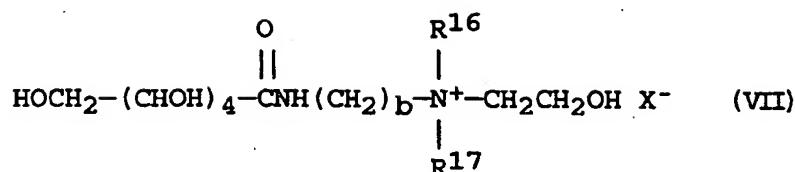
wherein Z² is an alkyl, preferably a C₁ - C₃ alkyl, more preferably methyl, and Z³ is a short chain hydroxyalkyl, preferably hydroxymethyl or hydroxyethyl, p and q independently are integers from 2 to 4, inclusive, preferably from 2 to 3, inclusive, more preferably 2, R¹¹ and R¹², independently, are substituted or unsubstituted hydrocarbyls, preferably C₁₂ - C₂₀ alkyl or alkenyl, and X is a salt forming anion as defined above;



wherein R¹³ is a hydrocarbyl, preferably a C₁ - C₃ alkyl, more preferably methyl, Z⁴ and Z⁵ are, independently, short chain hydrocarbyls, preferably C₂ - C₄ alkyl or alkenyl, more preferably ethyl, a is from 2 to about 40, preferably from about 7 to about 30, and X is a salt forming anion as defined above;



wherein R¹⁴ and R¹⁵, independently, are C₁ - C₃ alkyl, preferably methyl, Z⁶ is a C₁₂ - C₂₂ hydrocarbyl, alkyl carboxy or alkylamido, and A is a protein, preferably a collagen, keratin, milk protein, silk, soy protein, wheat protein, or hydrolyzed forms thereof; and X is a salt forming anion as defined above;



wherein b is 2 or 3, R¹⁶ and R¹⁷, independently are C₁ - C₃ hydrocarbyls preferably methyl, and X is a salt forming anion as defined above. Nonlimiting examples of hydrophilically substituted cationic surfactants useful include the materials having the following CTFA designations: quaternium-16, quaternium-61, quaternium-71, quaternium-79 hydrolyzed collagen, quaternium-79 hydrolyzed keratin, quaternium-79 hydrolyzed milk protein, quaternium-79 hydrolyzed silk, quaternium-79 hydrolyzed soy protein, and quaternium-79 hydrolyzed wheat protein. Highly preferred compounds include commercially available materials of the following tradenames; VARIQUAT K1215 and 638 from Witco Chemical, MACKPRO KLP, MACKPRO WLW, MACKPRO MLP, MACKPRO NSP, MACKPRO NLW, MACKPRO WWP, MACKPRO NLP, MACKPRO SLP from McIntyre, ETHOQUAD 18/25, ETHOQUAD O/12PG, ETHOQUAD C/25, ETHOQUAD S/25, and ETHODUOQUAD from Akzo, DEHYQUAT SP from Henkel, and ATLAS G265 from ICI Americas.

Salts of primary, secondary, and tertiary fatty amines are also suitable cationic surfactants. The alkyl groups of such amines preferably have from about 12 to about 22 carbon atoms, and can be substituted or unsubstituted. Particularly useful are amido substituted tertiary fatty amines. Such amines, useful herein, include stearamidopropyldimethylamine, stearamidopropyldiethylamine, stearamidoethyldiethylamine, stearamidoethyldimethylamine, palmitamidopropyldimethylamine, palmitamidopropyldiethylamine, palmitamidoethyldiethylamine, palmitamidoethyldimethylamine, behenamidopropyldimethylamine, behenamidopropyldiethylamine, behenamidoethyldiethylamine, behenamidoethyldimethylamine, arachidamidopropyldimethylamine, arachidamidopropyldiethylamine, arachidamidoethyldiethylamine, arachidamidoethyldimethylamine, diethylaminoethylstearamide. Also useful are dimethylstearamine, dimethylsoyamine, soyamine, myristylamine, tridecylamine, ethylstearylamine, N-tallowpropane diamine, ethoxylated (with 5 moles of ethylene oxide) stearylamine, dihydroxyethylstearylamine, and arachidylbehenylamine. These amines can also be used in combination with

acids such as L-glutamic acid, lactic acid, hydrochloric acid, malic acid, succinic acid, acetic acid, fumaric acid, tartaric acid, citric acid, L-glutamic hydrochloride, and mixtures thereof; more preferably L-glutamic acid, lactic acid, citric acid. Cationic amine surfactants included among those useful are disclosed in U.S. Patent 4,275,055, Nachtigal, et al., issued June 23, 1981, which is incorporated by reference herein in its entirety.

The cationic surfactants for use herein may also include a plurality of ammonium quaternary moieties or amino moieties, or a mixture thereof.

High Melting Point Compound

The compositions may comprise a high melting point compound having a melting point of at least about 25°C selected from the group consisting of fatty alcohols, fatty acids, fatty alcohol derivatives, fatty acid derivatives, hydrocarbons, steroids, and mixtures thereof. Without being bound by theory, it is believed that these high melting point compounds cover the hair surface and reduce friction, thereby resulting in providing smooth feel on the hair and ease of combing. It is understood by the artisan that the compounds disclosed in this section of the specification can in some instances fall into more than one classification, e.g., some fatty alcohol derivatives can also be classified as fatty acid derivatives. However, a given classification is not intended to be a limitation on that particular compound, but is done so for convenience of classification and nomenclature. Further, it is understood by the artisan that, depending on the number and position of double bonds, and length and position of the branches, certain compounds having certain required carbon atoms may have a melting point of less than about 25°C. Such compounds of low melting point are not intended to be included in this section. Nonlimiting examples of the high melting point compounds are found in International Cosmetic Ingredient Dictionary, Fifth Edition, 1993, and CTFA Cosmetic Ingredient Handbook, Second Edition, 1992, both of which are incorporated by reference herein in their entirety.

The fatty alcohols useful herein are those having from about 14 to about 30 carbon atoms, preferably from about 16 to about 22 carbon atoms. These fatty alcohols can be straight or branched chain alcohols and can be saturated or unsaturated. Nonlimiting examples of fatty alcohols include, cetyl alcohol, stearyl alcohol, behenyl alcohol, and mixtures thereof.

The fatty acids useful herein are those having from about 10 to about 30 carbon atoms, preferably from about 12 to about 22 carbon atoms, and more preferably from about 16 to about 22 carbon atoms. These fatty acids can be

straight or branched chain acids and can be saturated or unsaturated. Also included are diacids, triacids, and other multiple acids which meet the requirements herein. Also included herein are salts of these fatty acids. Nonlimiting examples of fatty acids include lauric acid, palmitic acid, stearic acid, behenic acid, sebacic acid, and mixtures thereof.

The fatty alcohol derivatives and fatty acid derivatives useful herein include alkyl ethers of fatty alcohols, alkoxyated fatty alcohols, alkyl ethers of alkoxyated fatty alcohols, esters of fatty alcohols, fatty acid esters of compounds having esterifiable hydroxy groups, hydroxy-substituted fatty acids, and mixtures thereof. Nonlimiting examples of fatty alcohol derivatives and fatty acid derivatives include materials such as methyl stearyl ether; the ceteth series of compounds such as ceteth-1 through ceteth-45, which are ethylene glycol ethers of cetyl alcohol, wherein the numeric designation indicates the number of ethylene glycol moieties present; the steareth series of compounds such as steareth-1 through 10, which are ethylene glycol ethers of steareth alcohol, wherein the numeric designation indicates the number of ethylene glycol moieties present; cetareth 1 through cetareth-10, which are the ethylene glycol ethers of cetareth alcohol, i.e. a mixture of fatty alcohols containing predominantly cetyl and stearyl alcohol, wherein the numeric designation indicates the number of ethylene glycol moieties present; C₁-C₃₀ alkyl ethers of the ceteth, steareth, and cetareth compounds just described; polyoxyethylene ethers of behenyl alcohol; ethyl stearate, cetyl stearate, cetyl palmitate, stearyl stearate, myristyl myristate, polyoxyethylene cetyl ether stearate, polyoxyethylene stearyl ether stearate, polyoxyethylene lauryl ether stearate, ethyleneglycol monostearate, polyoxyethylene monostearate, polyoxyethylene distearate, propyleneglycol monostearate, propyleneglycol distearate, trimethylolpropane distearate, sorbitan stearate, polyglyceryl stearate, glyceryl monostearate, glyceryl distearate, glyceryl tristearate, and mixtures thereof.

Hydrocarbons useful herein include compounds having at least about 20 carbons.

Steroids useful herein include compounds such as cholesterol.

High melting point compounds of a single compound of high purity are preferred. Single compounds of pure fatty alcohols selected from the group of pure cetyl alcohol, stearyl alcohol, and behenyl alcohol are highly preferred. By "pure" herein, what is meant is that the compound has a purity of at least about 90%, preferably at least about 95%. These single compounds of high purity

provide good rinsability from the hair when the consumer rinses off the composition.

Commercially available high melting point compounds useful herein include: cetyl alcohol, stearyl alcohol, and behenyl alcohol having tradenames
5 KONOL series available from New Japan Chemical (Osaka, Japan), and NAA series available from NOF (Tokyo, Japan); pure behenyl alcohol having tradename 1-DOCOSANOL available from WAKO (Osaka, Japan), various fatty acids having tradenames NEO-FAT available from Akzo (Chicago Illinois, USA), HYSTRENE available from Witco Corp. (Dublin Ohio, USA), and DERMA
10 available from Vevy (Genova, Italy); and cholesterol having tradename NIKKOL AGUASOME LA available from Nikko.

Oily Compound

The compositions comprise an oily compound having a melting point of not more than about 25°C selected from the group consisting of a first oily
15 compound, a second oily compound, and mixtures thereof. The oily compounds useful herein may be volatile or nonvolatile. Without being bound by theory, it is believed that, the oily compounds may penetrate the hair to modify the hydroxy bonds of the hair, thereby resulting in providing softness and flexibility to the hair. The oily compound may comprise either the first oily compound or the second
20 oily compound as described herein. Preferably, a mixture of the first oily compound and the second oily compound is used. The oily compounds of this section are to be distinguished from the high melting point compounds described above. Nonlimiting examples of the oily compounds are found in International Cosmetic Ingredient Dictionary, Fifth Edition, 1993, and CTFA Cosmetic
25 Ingredient Handbook, Second Edition, 1992, both of which are incorporated by reference herein in their entirety.

First Oily Compound

The fatty alcohols useful herein include those having from about 10 to about 30 carbon atoms, preferably from about 12 to about 22 carbon atoms, and
30 more preferably from about 16 to about 22 carbon atoms. These fatty alcohols can be straight or branched chain alcohols and can be saturated or unsaturated alcohols, preferably unsaturated alcohols. Nonlimiting examples of these compounds include oleyl alcohol, palmitoleic alcohol, isostearyl alcohol, isocetyl alcohol, undecanol, octyl dodecanol, octyl decanol, octyl alcohol, caprylic alcohol,
35 decyl alcohol and lauryl alcohol.

The fatty acids useful herein include those having from about 10 to about 30 carbon atoms, preferably from about 12 to about 22 carbon atoms, and more preferably from about 16 to about 22 carbon atoms. These fatty acids can be straight or branched chain acids and can be saturated or unsaturated. Suitable fatty acids include, for example, oleic acid, linoleic acid, isostearic acid, linolenic acid, ethyl linolenic acid, ethyl linolenic acid, arachidonic acid, and ricinolic acid.

The fatty acid derivatives and fatty alcohol derivatives are defined herein to include, for example, esters of fatty alcohols, alkoxyated fatty alcohols, alkyl ethers of fatty alcohols, alkyl ethers of alkoxyated fatty alcohols, and mixtures thereof. Nonlimiting examples of fatty acid derivatives and fatty alcohol derivatives, include, for example, methyl linoleate, ethyl linoleate, isopropyl linoleate, isodecyl oleate, isopropyl oleate, ethyl oleate, octyldodecyl oleate, oleyl oleate, decyl oleate, butyl oleate, methyl oleate, octyldodecyl stearate, octyldodecyl isostearate, octyldodecyl isopalmitate, octyl isopelargonate, octyl pelargonate, hexyl isostearate, isopropyl isostearate, isodecyl isononanoate, Oleth-2, pentaerythritol tetraoleate, pentaerythritol tetraisostearate, trimethylolpropane trioleate, and trimethylolpropane triisostearate.

Commercially available first oily compounds useful herein include: oleyl alcohol with tradename UNJECOL 90BHR available from New Japan Chemical, pentaerythritol tetraisostearate and trimethylolpropane triisostearate with tradenames KAKPTI and KAKTTI available from Kokyu Alcohol (Chiba, Japan), pentaerythritol tetraoleate having the same tradename as the compound name available from New Japan Chemical, trimethylolpropane trioleate with a tradename ENUJERUBU TP3SO available from New Japan Chemical, various liquid esters with tradenames SCHERCEMOL series available from Scher, and hexyl isostearate with a tradename HIS and isopropyl isostearate having a tradename ZPIS available from Kokyu Alcohol.

Second Oily Compound

The second oily compounds useful herein include straight chain, cyclic, and branched chain hydrocarbons which can be either saturated or unsaturated, so long as they have a melting point of not more than about 25°C. These hydrocarbons have from about 12 to about 40 carbon atoms, preferably from about 12 to about 30 carbon atoms, and preferably from about 12 to about 22 carbon atoms. Also encompassed herein are polymeric hydrocarbons of alkenyl monomers, such as polymers of C₂₋₆ alkenyl monomers. These polymers can be straight or branched chain polymers. The straight chain polymers will typically

be relatively short in length, having a total number of carbon atoms as described above. The branched chain polymers can have substantially higher chain lengths. The number average molecular weight of such materials can vary widely, but will typically be up to about 500, preferably from about 200 to about 5 400, and more preferably from about 300 to about 350. Also useful herein are the various grades of mineral oils. Mineral oils are liquid mixtures of hydrocarbons that are obtained from petroleum. Specific examples of suitable hydrocarbon materials include paraffin oil, mineral oil, dodecane, isododecane, hexadecane, isohexadecane, eicosene, isoeicosene, tridecane, tetradecane, 10 polybutene, polyisobutene, and mixtures thereof. Preferred for use herein are hydrocarbons selected from the group consisting of mineral oil, isododecane, isohexadecane, polybutene, polyisobutene, and mixtures thereof.

Commercially available second oily compounds useful herein include isododecane, isohexadecane, and isoeicosene with tradenames PERMETHYL 15 99A, PERMETHYL 101A, and PERMETHYL 1082, available from Presperse (South Plainfield New Jersey, USA), a copolymer of isobutene and normal butene with tradenames INDOPOL-H-100 available from Amoco Chemicals (Chicago Illinois, USA), mineral oil with tradename BENOL available from Witco, isoparaffin with tradename ISOPAR from Exxon Chemical Co. (Houston Texas, 20 USA), α -olefin oligomer with tradename PURESYN 6 from Mobil Chemical Co., and trimethylolpropane tricaprilate/tricaprate with tradename MOBIL ESTER P43 from Mobil Chemical Co.

Cationic Polymers

As used herein, the term "polymer" shall include materials whether made 25 by polymerization of one type of monomer or made by two (i.e., copolymers) or more types of monomers.

Preferably, the cationic polymer is a water-soluble cationic polymer. By "water soluble" cationic polymer, what is meant is a polymer which is sufficiently soluble in water to form a substantially clear solution to the naked eye at a 30 concentration of 0.1% in water (distilled or equivalent) at 25°C. The preferred polymer will be sufficiently soluble to form a substantially clear solution at 0.5% concentration, more preferably at 1.0% concentration.

The cationic polymers hereof will generally have a weight average molecular weight which is at least about 5,000, typically at least about 10,000, 35 and is less than about 10 million. Preferably, the molecular weight is from about 100,000 to about 2 million. The cationic polymers will generally have cationic

nitrogen-containing moieties such as quaternary ammonium or cationic amino moieties, and mixtures thereof.

The cationic charge density is preferably at least about 0.1 meq/gram, more preferably at least about 1.5 meq/gram, even more preferably at least about 1.1 meq/gram, still more preferably at least about 1.2 meq/gram. Cationic charge density of the cationic polymer can be determined according to the Kjeldahl Method. Those skilled in the art will recognize that the charge density of amino-containing polymers may vary depending upon pH and the isoelectric point of the amino groups. The charge density should be within the above limits at the pH of intended use.

Any anionic counterions can be utilized for the cationic polymers so long as the water solubility criteria is met. Suitable counterions include halides (e.g., Cl, Br, I, or F, preferably Cl, Br, or I), sulfate, and methylsulfate. Others can also be used, as this list is not exclusive.

The cationic nitrogen-containing moiety will be present generally as a substituent, on a fraction of the total monomer units of the cationic hair conditioning polymers. Thus, the cationic polymer can comprise copolymers, terpolymers, etc. of quaternary ammonium or cationic amine-substituted monomer units and other non-cationic units referred to herein as spacer monomer units. Such polymers are known in the art, and a variety can be found in the CTFA Cosmetic Ingredient Dictionary, 3rd edition, edited by Estrin, Crosley, and Haynes, (The Cosmetic, Toiletry, and Fragrance Association, Inc., Washington, D.C., 1982).

Suitable cationic polymers include, for example, copolymers of vinyl monomers having cationic amine or quaternary ammonium functionalities with water soluble spacer monomers such as acrylamide, methacrylamide, alkyl and dialkyl acrylamides, alkyl and dialkyl methacrylamides, alkyl acrylate, alkyl methacrylate, vinyl caprolactone, and vinyl pyrrolidone. The alkyl and dialkyl substituted monomers preferably have C₁ - C₇ alkyl groups, more preferably C₁ - C₃ alkyl groups. Other suitable spacer monomers include vinyl esters, vinyl alcohol (made by hydrolysis of polyvinyl acetate), maleic anhydride, propylene glycol, and ethylene glycol.

The cationic amines can be primary, secondary, or tertiary amines, depending upon the particular species and the pH of the composition. In general, secondary and tertiary amines, especially tertiary amines, are preferred.

Amine-substituted vinyl monomers can be polymerized in the amine form, and then optionally can be converted to ammonium by a quaternization reaction. Amines can also be similarly quaternized subsequent to formation of the polymer. For example, tertiary amine functionalities can be quaternized by
5 reaction with a salt of the formula R'X wherein R' is a short chain alkyl, preferably a C₁ - C₇ alkyl, more preferably a C₁ - C₃ alkyl, and X is an anion which forms a water soluble salt with the quaternized ammonium.

Suitable cationic amino and quaternary ammonium monomers include, for example, vinyl compounds substituted with dialkylaminoalkyl acrylate,
10 dialkylaminoalkyl methacrylate, monoalkylaminoalkyl acrylate, monoalkylaminoalkyl methacrylate, trialkyl methacryloxyalkyl ammonium salt, trialkyl acryloxyalkyl ammonium salt, diallyl quaternary ammonium salts, and vinyl quaternary ammonium monomers having cyclic cationic nitrogen-containing rings such as pyridinium, imidazolium, and quaternized pyrrolidone, e.g., alkyl vinyl
15 imidazolium, alkyl vinyl pyridinium, alkyl vinyl pyrrolidone salts. The alkyl portions of these monomers are preferably lower alkyls such as the C₁ - C₃ alkyls, more preferably C₁ and C₂ alkyls. Suitable amine-substituted vinyl monomers for use herein include dialkylaminoalkyl acrylate, dialkylaminoalkyl methacrylate, dialkylaminoalkyl acrylamide, and dialkylaminoalkyl methacrylamide, wherein the
20 alkyl groups are preferably C₁ - C₇ hydrocarbyls, more preferably C₁ - C₃ alkyls.

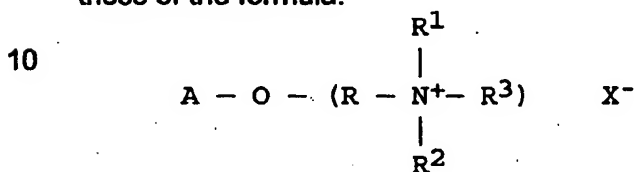
The cationic polymers hereof can comprise mixtures of monomer units derived from amine- and/or quaternary ammonium-substituted monomer and/or compatible spacer monomers.

25 Suitable cationic hair conditioning polymers include, for example: copolymers of 1-vinyl-2-pyrrolidone and 1-vinyl-3-methylimidazolium salt (e.g., chloride salt) (referred to in the industry by the Cosmetic, Toiletry, and Fragrance Association, "CTFA", as Polyquaternium-16), such as those commercially available from BASF Wyandotte Corp. (Parsippany, NJ, USA) under the
30 LUVIQUAT tradename (e.g., LUVIQUAT FC 370); copolymers of 1-vinyl-2-pyrrolidone and dimethylaminoethyl methacrylate (referred to in the industry by CTFA as Polyquaternium-11) such as those commercially available from Gaf Corporation (Wayne, NJ, USA) under the GAFQUAT tradename (e.g., GAFQUAT 755N); cationic diallyl quaternary ammonium-containing polymers, including, for
35 example, dimethyldiallylammonium chloride homopolymer and copolymers of acrylamide and dimethyldiallylammonium chloride, referred to in the industry

(CTFA) as Polyquaternium 6 and Polyquaternium 7, respectively; and mineral acid salts of amino-alkyl esters of homo- and co-polymers of unsaturated carboxylic acids having from 3 to 5 carbon atoms, as described in U.S. Patent 4,009,256, incorporated herein by reference.

- 5 Other cationic polymers that can be used include polysaccharide polymers, such as cationic cellulose derivatives and cationic starch derivatives.

Cationic polysaccharide polymer materials suitable for use herein include those of the formula:



- 15 wherein: A is an anhydroglucose residual group, such as a starch or cellulose anhydroglucose residual, R is an alkylene oxyalkylene, polyoxyalkylene, or hydroxyalkylene group, or combination thereof, R¹, R², and R³ independently are alkyl, aryl, alkylaryl, arylalkyl, alkoxyalkyl, or alkoxyaryl groups, each group containing up to about 18 carbon atoms, and the total number of carbon atoms
20 for each cationic moiety (i.e., the sum of carbon atoms in R¹, R² and R³) preferably being about 20 or less, and X is an anionic counterion, as previously described.

- Cationic cellulose is available from Amerchol Corp. (Edison, NJ, USA) in their Polymer JR® and LR® series of polymers, as salts of hydroxyethyl cellulose
25 reacted with trimethyl ammonium substituted epoxide, referred to in the industry (CTFA) as Polyquaternium 10. Another type of cationic cellulose includes the polymeric quaternary ammonium salts of hydroxyethyl cellulose reacted with lauryl dimethyl ammonium-substituted epoxide, referred to in the industry (CTFA) as Polyquaternium 24. These materials are available from Amerchol Corp.
30 (Edison, NJ, USA) under the tradename Polymer LM-200®.

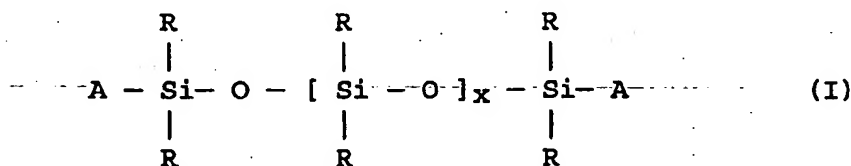
- Other cationic polymers that can be used include cationic guar gum derivatives, such as guar hydroxypropyltrimonium chloride (commercially available from Celanese Corp. in their Jaguar R series). Other materials include quaternary nitrogen-containing cellulose ethers (e.g., as described in U.S. Patent
35 3,962,418, incorporated herein by reference), and copolymers of etherified cellulose and starch (e.g., as described in U.S. Patent 3,958,581, incorporated herein by reference.)

Silicone Compounds

The conditioning agents useful herein include silicone compounds. The silicone compounds hereof can include volatile soluble or insoluble, or nonvolatile soluble or insoluble silicone conditioning agents. By soluble what is meant is that the silicone compound is miscible with the carrier of the composition so as to form part of the same phase. By insoluble what is meant is that the silicone forms a separate, discontinuous phase from the carrier, such as in the form of an emulsion or a suspension of droplets of the silicone.

Suitable silicone fluids include polyalkyl siloxanes, polyaryl siloxanes, polyalkylaryl siloxanes, polyether siloxane copolymers, and mixtures thereof. Other nonvolatile silicone compounds having hair conditioning properties can also be used.

The silicone compounds herein also include polyalkyl or polyaryl siloxanes with the following structure (I)



wherein R is alkyl or aryl, and x is an integer from about 7 to about 8,000. "A" represents groups which block the ends of the silicone chains. The alkyl or aryl groups substituted on the siloxane chain (R) or at the ends of the siloxane chains (A) can have any structure as long as the resulting silicone remains fluid at room temperature, is dispersible, is neither irritating, toxic nor otherwise harmful when applied to the hair, is compatible with the other components of the composition, is chemically stable under normal use and storage conditions, and is capable of being deposited on and conditions the hair. Suitable A groups include hydroxy, methyl, methoxy, ethoxy, propoxy, and aryloxy. The two R groups on the silicon atom may represent the same group or different groups. Preferably, the two R groups represent the same group. Suitable R groups include methyl, ethyl, propyl, phenyl, methylphenyl and phenylmethyl. The preferred silicone compounds are polydimethylsiloxane, polydiethylsiloxane, and polymethylphenylsiloxane. Polydimethylsiloxane, which is also known as dimethicone, is especially preferred. The polyalkylsiloxanes that can be used include, for example, polydimethylsiloxanes. These silicone compounds are

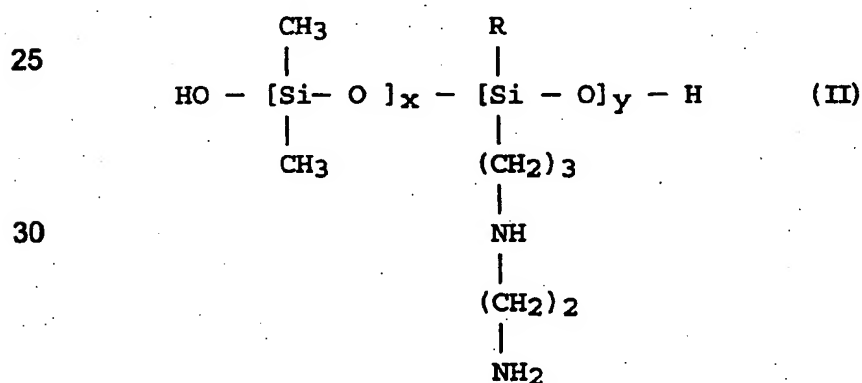
available, for example, from the General Electric Company in their ViscasilR and SF 96 series, and from Dow Corning in their Dow Corning 200 series.

Polyalkylaryl siloxane fluids can also be used and include, for example, polymethylphenylsiloxanes. These siloxanes are available, for example, from the
 5 General Electric Company as SF 1075 methyl phenyl fluid or from Dow Corning as 556 Cosmetic Grade Fluid.

Especially preferred, for enhancing the shine characteristics of hair, are highly arylated silicone compounds, such as highly phenylated polyethyl silicone having refractive index of about 1.46 or higher, especially about 1.52 or higher.
 10 When these high refractive index silicone compounds are used, they should be mixed with a spreading agent, such as a surfactant or a silicone resin, as described below to decrease the surface tension and enhance the film forming ability of the material.

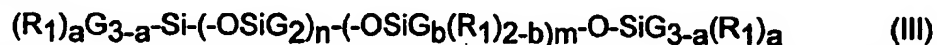
The silicone compounds that can be used include, for example, a
 15 polypropylene oxide modified polydimethylsiloxane although ethylene oxide or mixtures of ethylene oxide and propylene oxide can also be used. The ethylene oxide and polypropylene oxide level should be sufficiently low so as not to interfere with the dispersibility characteristics of the silicone. These material are also known as dimethicone copolyols.

20 Other silicone compounds include amino substituted materials. Suitable alkylamino substituted silicone compounds include those represented by the following structure (II)

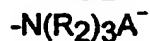
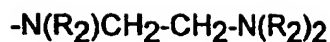


wherein R is CH₃ or OH, x and y are integers which depend on the molecular weight, the average molecular weight being approximately between 5,000 and 10,000. This polymer is also known as "amodimethicone".

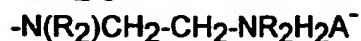
Suitable amino substituted silicone fluids include those represented by the formula (III)



- 5 in which G is chosen from the group consisting of hydrogen, phenyl, OH, C₁-C₈ alkyl and preferably methyl; a denotes 0 or an integer from 1 to 3, and preferably equals 0; b denotes 0 or 1 and preferably equals 1; the sum n+m is a number from 1 to 2,000 and preferably from 50 to 150, n being able to denote a number from 0 to 1,999 and preferably from 49 to 149 and m being able to denote an integer from 1 to 2,000 and preferably from 1 to 10; R₁ is a monovalent radical
- 10 of formula C_qH_{2q}L in which q is an integer from 2 to 8 and L is chosen from the groups

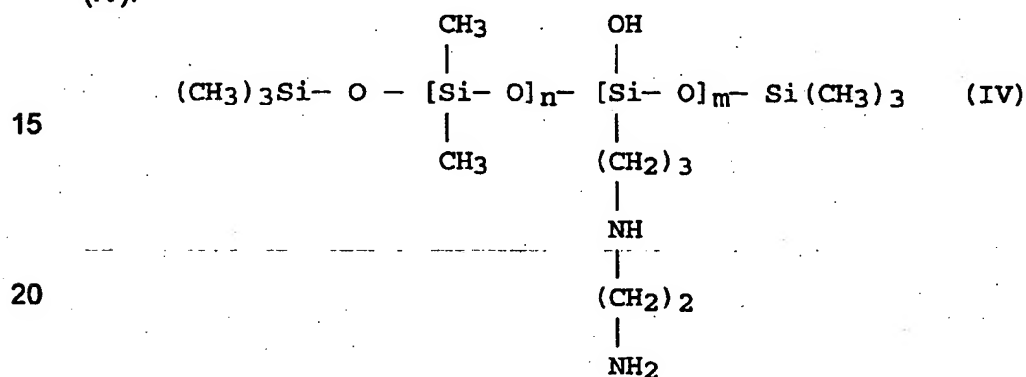


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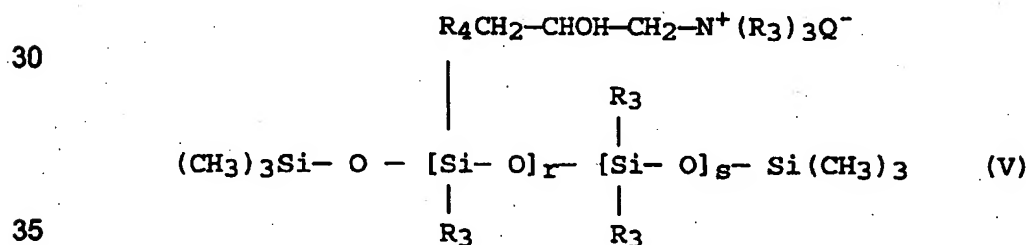
in which R_2 is chosen from the group consisting of hydrogen, phenyl, benzyl, a saturated hydrocarbon radical, preferably an alkyl radical containing from 1 to 20 carbon atoms, and A^- denotes a halide ion.

10 An especially preferred amino substituted silicone corresponding to formula (III) is the polymer known as "trimethylsilylamodimethicone", of formula (IV):



25 In this formula n and m are selected depending on the exact molecular weight of the compound desired.

Other amino substituted silicone polymers which can be used are represented by the formula (V):



where R^3 denotes a monovalent hydrocarbon radical having from 1 to 18 carbon atoms, preferably an alkyl or alkenyl radical such as methyl; R_4 denotes a hydrocarbon radical, preferably a $C_1 - C_{18}$ alkylene radical or a $C_1 - C_{18}$, and
 40 more preferably $C_1 - C_8$, alkyleneoxy radical; Q^- is a halide ion, preferably

chloride; r denotes an average statistical value from 2 to 20, preferably from 2 to 8; s denotes an average statistical value from 20 to 200, and preferably from 20 to 50. A preferred polymer of this class is available from Union Carbide under the name "UCAR SILICONE ALE 56."

5 References disclosing suitable nonvolatile dispersed silicone compounds include U.S. Patent No. 2,826,551, to Geen; U.S. Patent No. 3,964,500, to Drakoff, issued June 22, 1976; U.S. Patent No. 4,364,837, to Pader; and British
10 Patent No. 849,433, to Woolston, all of which are incorporated herein by reference in their entirety. Also incorporated herein by reference in its entirety is "Silicon Compounds" distributed by Petrarch Systems, Inc., 1984. This reference provides an extensive, though not exclusive, listing of suitable silicone compounds.

 Another nonvolatile dispersed silicone that can be especially useful is a
15 silicone gum. The term "silicone gum", as used herein, means a polyorganosiloxane material having a viscosity at 25°C of greater than or equal to 1,000,000 centistokes. It is recognized that the silicone gums described herein can also have some overlap with the above-disclosed silicone compounds. This overlap is not intended as a limitation on any of these materials. Silicone gums are described by Petrarch, and others including U.S. Patent No. 4,152,416, to
20 Spitzer et al., issued May 1, 1979 and Noll, Walter, Chemistry and Technology of Silicones, New York: Academic Press 1968. Also describing silicone gums are General Electric Silicone Rubber Product Data Sheets SE 30, SE 33, SE 54 and SE 76. All of these described references are incorporated herein by reference in their entirety. The "silicone gums" will typically have a mass molecular weight in
25 excess of about 200,000, generally between about 200,000 and about 1,000,000. Specific examples include polydimethylsiloxane, poly(dimethylsiloxane methylvinylsiloxane) copolymer, poly(dimethylsiloxane diphenylsiloxane methylvinylsiloxane) copolymer and mixtures thereof.

 Also useful are silicone resins, which are highly crosslinked polymeric
30 siloxane systems. The crosslinking is introduced through the incorporation of tri-functional and tetra-functional silanes with mono-functional or di-functional, or both, silanes during manufacture of the silicone resin. As is well understood in the art, the degree of crosslinking that is required in order to result in a silicone resin will vary according to the specific silane units incorporated into the silicone
35 resin. In general, silicone materials which have a sufficient level of trifunctional and tetrafunctional siloxane monomer units, and hence, a sufficient level of

crosslinking, such that they dry down to a rigid, or hard, film are considered to be silicone resins. The ratio of oxygen atoms to silicon atoms is indicative of the level of crosslinking in a particular silicone material. Silicone materials which have at least about 1.1 oxygen atoms per silicon atom will generally be silicone resins herein. Preferably, the ratio of oxygen:silicon atoms is at least about 1.2:1.0. Silanes used in the manufacture of silicone resins include monomethyl-, dimethyl-, trimethyl-, monophenyl-, diphenyl-, methylphenyl-, monovinyl-, and methylvinylchlorosilanes, and tetrachlorosilane, with the methyl substituted silanes being most commonly utilized. Preferred resins are offered by General Electric as GE SS4230 and SS4267. Commercially available silicone resins will generally be supplied in a dissolved form in a low viscosity volatile or nonvolatile silicone fluid. The silicone resins for use herein should be supplied and incorporated into the present compositions in such dissolved form, as will be readily apparent to those skilled in the art. Without being bound by theory, it is believed that the silicone resins can enhance deposition of other silicone compounds on the hair and can enhance the glossiness of hair with high refractive index volumes.

Other useful silicone resins are silicone resin powders such as the material given the CTFA designation polymethylsilsequioxane, which is commercially available as TospearlTM from Toshiba Silicones.

The method of manufacturing these silicone compounds, can be found in Encyclopedia of Polymer Science and Engineering, Volume 15, Second Edition, pp 204-308, John Wiley & Sons, Inc., 1989, which is incorporated herein by reference in its entirety.

Silicone materials and silicone resins in particular, can conveniently be identified according to a shorthand nomenclature system well known to those skilled in the art as the "MDTQ" nomenclature. Under this system, the silicone is described according to the presence of various siloxane monomer units which make up the silicone. Briefly, the symbol M denotes the mono-functional unit $(\text{CH}_3)_3\text{SiO}_{0.5}$; D denotes the difunctional unit $(\text{CH}_3)_2\text{SiO}$; T denotes the trifunctional unit $(\text{CH}_3)\text{SiO}_{1.5}$; and Q denotes the quadri- or tetra-functional unit SiO_2 . Primes of the unit symbols, e.g., M', D', T', and Q' denote substituents other than methyl, and must be specifically defined for each occurrence. Typical alternate substituents include groups such as vinyl, phenyl, amino, hydroxyl, etc. The molar ratios of the various units, either in terms of subscripts to the symbols indicating the total number of each type of unit in the silicone, or an average

thereof, or as specifically indicated ratios in combination with molecular weight, complete the description of the silicone material under the MDTQ system. Higher relative molar amounts of T, Q, T' and/or Q' to D, D', M and/or M' in a silicone resin is indicative of higher levels of crosslinking. As discussed before, however, the overall level of crosslinking can also be indicated by the oxygen to silicon ratio.

The silicone resins for use herein which are preferred are MQ, MT, MTQ, MQ and MDTQ resins. Thus, the preferred silicone substituent is methyl. Especially preferred are MQ resins wherein the M:Q ratio is from about 0.5:1.0 to about 1.5:1.0 and the average molecular weight of the resin is from about 1000 to about 10,000.

Nonionic Polymer

Nonionic polymers useful herein include cellulose derivatives, hydrophobically modified cellulose derivatives, ethylene oxide polymers, and ethylene oxide/propylene oxide based polymers. Suitable nonionic polymers are cellulose derivatives including methylcellulose with tradename BENECEL, hydroxyethyl cellulose with tradename NATROSOL, hydroxypropyl cellulose with tradename KLUCEL, cetyl hydroxyethyl cellulose with tradename POLYSURF 67, all supplied by Hercules. Other suitable nonionic polymers are ethylene oxide and/or propylene oxide based polymers with tradenames CARBOWAX PEGs, POLYOX WASRs, and UCON FLUIDS, all supplied by Amerchol.

Polyalkylene Glycols

These compounds are particularly useful for compositions which are designed to impart a soft, moist feeling to the hair. When present, the polyalkylene glycol is typically used at a level from about 0.025% to about 1.5%, preferably from about 0.05% to about 1%, and more preferably from about 0.1% to about 0.5% of the compositions.

The polyalkylene glycols are characterized by the general formula:



wherein R is selected from the group consisting of H, methyl, and mixtures thereof. When R is H, these materials are polymers of ethylene oxide, which are also known as polyethylene oxides, polyoxyethylenes, and polyethylene glycols. When R is methyl, these materials are polymers of propylene oxide, which are also known as polypropylene oxides, polyoxypropylenes, and polypropylene

glycols. When R is methyl, it is also understood that various positional isomers of the resulting polymers can exist.

In the above structure, n has an average value of from about 1500 to about 25,000, preferably from about 2500 to about 20,000, and more preferably from about 3500 to about 15,000.

5 Polyethylene glycol polymers useful herein are PEG-2M wherein R equals H and n has an average value of about 2,000 (PEG-2M is also known as Polyox WSR® N-10, which is available from Union Carbide and as PEG-2,000); PEG-5M wherein R equals H and n has an average value of about 5,000 (PEG-5M is also
10 known as Polyox WSR® N-35 and Polyox WSR® N-80, both available from Union Carbide and as PEG-5,000 and Polyethylene Glycol 300,000); PEG-7M wherein R equals H and n has an average value of about 7,000 (PEG-7M is also known as Polyox WSR® N-750 available from Union Carbide); PEG-9M wherein R equals H and n has an average value of about 9,000 (PEG 9-M is also known
15 as Polyox WSR® N-3333 available from Union Carbide); and PEG-14 M wherein R equals H and n has an average value of about 14,000 (PEG-14M is also known as Polyox WSR® N-3000 available from Union Carbide).

Other useful polymers include the polypropylene glycols and mixed polyethylene/polypropylene glycols.

20 FIXATIVE POLYMERS

Compositions herein may include a fixative polymer. The fixative polymers useful herein are those which provide a styling or setting benefit to the hair, including amphoteric fixative polymers, anionic fixative polymers, cationic fixative polymers, nonionic fixative polymers, and silicone grafted copolymers.
25 The fixative polymers in this section can be distinguished from the conditioning agent polymers mentioned above and the suspending agent polymers as mentioned below, in that they have a film-forming characteristic, and that the 3% aqueous solution of fixative polymers herein provide a viscosity of no more than about 2,000 cps.

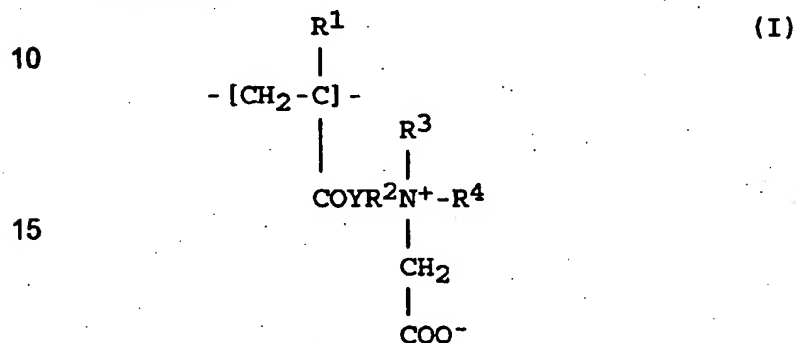
30 The fixative polymers, when present, are preferably included at a level of from about 0.01% to about 10% by weight of the composition. When two or more fixative polymers are used, each fixative polymer does not exceed about 5% by weight of the composition.

Amphoteric Fixative Polymer

The amphoteric fixative polymers useful herein include betainized amphoteric fixative polymers as described in (1) and (2) below, and non-betainized amphoteric fixative polymers as described in (3) and (4) below.

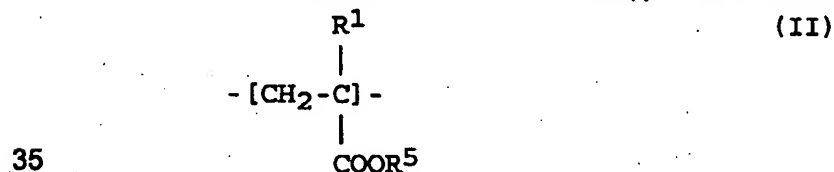
5 Betainized Amphoteric Fixative Polymers

(1) Useful herein are polymers of betainized dialkylaminoalkyl (meth)acrylate or dialkylaminoalkyl (meth)acrylamide containing at least units of the formula:



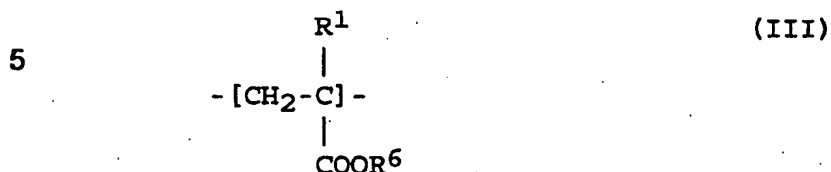
wherein R^1 denotes a hydrogen atom or a methyl group, R^2 denotes an alkylene group having 1 to 4 carbon atoms, Y denotes O or -NH- and R^3 and R^4 independently of one another denote hydrogen or alkyl having 1 to 4 carbon atoms, and one cationic derivative consisting of a cationic surfactant containing at least one nitrogen atom joined to one or more fatty chains and optionally quaternised, or consisting of a cationic polymer of the polyamine, polyaminopolyamide or poly-(quaternary ammonium) type, the amine or ammonium groups forming part of the polymer chain or being joined thereto. These polymers usually have a molecular weight of 500 to 2,000,000.

The amphoteric polymers containing units corresponding to the above formula (I) are generally in the form of copolymers which contain, in addition to the units of the above mentioned formula (I), at least units of the formula:



wherein R^1 is as defined above and R^5 represents an alkyl or alkenyl radical having from 4 to 24 carbon atoms or a cycloalkyl radical having from 4 to 24 carbon atoms.

It is also possible to use terpolymers, tetrapolymers or pentapolymers which contain, in addition to the units (I) and (II) defined above, units of the formula:

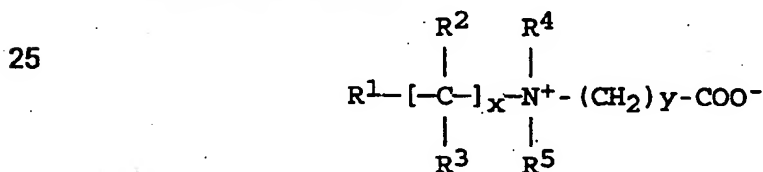


10 wherein R⁶ preferably denotes an alkyl or alkenyl group having 1 to 3 carbon atoms and R¹ is as defined above.

The units of the formula (I) are preferably present in an amount of 25 to 45% by weight, units of the formula (II) are preferably present in an amount of 5 to 65% by weight, and units of the formula (III) are preferably present in an amount up to 50% by weight, relative to the total weight of the polymer.

15 A particularly preferred polymer is the copolymer containing units of the formulae (I), (II) and (III) in which Y denotes an oxygen atom, R² denotes the group -C₂H₄-, R¹, R³ and R⁴ denote methyl, R⁵ denotes an alkyl group having 4 to 18 carbon atoms and R⁶ denotes an alkyl group having 1 to 3 carbon atoms. The average molecular weight of this polymer is preferably from 50,000 to 100,000. This polymer is sold under the trademark "Yukaformer" or "Diaformer" supplied by Mitsubishi Chemical Corporation.

(2) Useful herein are the betainized polymers containing zwitterionic units derived from the formula:



30 wherein R¹ denotes a polymerisable unsaturated group, such as an acrylate, methacrylate, acrylamide or methacrylamide group, x and y independently represent an integer from 1 to 3, R² and R³ independently represent hydrogen, methyl, ethyl or propyl, and R⁴ and R⁵ independently represent a hydrogen atom or an alkyl radical such that the sum of the carbon atoms in R⁴ and R⁵ does not exceed 10.

35 Highly preferred betainized amphoteric polymers include commercially available material such as YUKAFORMER SM, YUKAFORMER FH, YUKAFORMER 301, YUKAFORMER 204WL, YUKAFORMER 510,

YUKAFORMER M-75, YUKAFORMER R250S, Diaformer Z-SM, and Diaformer Z-W supplied by Mitsubishi Chemical Corporation.

Non-betainized Amphoteric Fixative Polymers

(3) Useful herein are the non-betainized amphoteric polymers resulting from the copolymerisation of a vinyl monomer carrying at least one carboxyl group, such as acrylic acid, methacrylic acid, maleic acid, itaconic acid, fumaric acid, crotonic acid, or alphachloroacrylic acid, and a basic monomer which is a substituted vinyl compound containing at least one basic nitrogen atom, such as dialkylaminoalkyl methacrylates and acrylates and dialkylaminoalkylmethacrylamides and -acrylamides.

(4) Useful herein are the non-betainized amphoteric polymers containing units derived from

- i) at least one monomer chosen from amongst acrylamides or methacrylamides substituted on the nitrogen by an alkyl radical,
- ii) at least one acid comonomer containing one or more reactive carboxyl groups, and
- iii) at least one basic comonomer, such as esters, with primary, secondary and tertiary amine substituents and quarternary ammonium substituents, of acrylic and methacrylic acids, and the product resulting from the quaternisation of dimethylaminoethyl methacrylate with dimethyl or diethyl sulphate.

The N-substituted acrylamides or methacrylamides which are most particularly preferred are the groups in which the alkyl radicals contain from 2 to 12 carbon atoms, especially N-ethylacrylamide, N-tert.-butylacrylamide, N-tert.-octylacrylamide, N-octylacrylamide, N-decylacrylamide and N-dodecylacrylamide and also the corresponding methacrylamides. The acid comonomers are chosen more particularly from amongst acrylic, methacrylic, crotonic, itaconic, maleic and fumaric acids and also the alkyl monoesters of maleic acid or fumaric acid in which alkyl has 1 to 4 carbon atoms.

The preferred basic comonomers are aminoethyl, butylaminoethyl, N,N'-dimethylaminoethyl and N-tert.-butylaminoethyl methacrylates.

Highly preferred non-betainized amphoteric polymers include commercially available material such as octylacrylamine/acrylates/butylaminoethyl methacrylate copolymers with the tradenames AMPHOMER, AMPHOMER SH701, AMPHOMER 28-4910,

AMPHOMER LV71, and AMPHOMER LV47 supplied by National Starch & Chemical.

Cationic Fixative Polymer

The cationic fixative polymers useful herein are:

- 5 (1) Vinylpyrrolidone / quaternized dialkylaminoalkyl acrylate or methacrylate copolymers such as those sold under the tradename Gafquat 734 and 755N by the Gaf Corp.

(2) Cellulose ether derivatives containing quaternary ammonium groups.

(3) Cationic polysaccharides.

- 10 (4) Cationic polymers chosen from the group comprising:

i) polymers containing units of the formula:



- wherein A denotes a radical containing two amino groups, preferably a piperazinyl radical, and Z¹ and Z² independently denote a divalent radical which is a straight-chain or branched-chain alkylene radical which contains up to about 7 carbon atoms in the main chain, is unsubstituted or substituted by one or more hydroxyl groups and can also contain one or more oxygen, nitrogen and sulphur atoms and 1 to 3 aromatic and/or heterocyclic rings, the oxygen, nitrogen and sulphur atoms generally being present in the form of an ether or thioether, sulphoxide, sulphone, sulphonium, amine, alkylamine, alkenylamine, benzylamine, amine oxide, quaternary ammonium, amide, imide, alcohol, ester and/or urethane group;
- 15 20

ii) polymers containing units of the formula:



- 25 wherein A denotes a radical containing two amino groups, preferably a piperazinyl radical, and Z¹ denotes the symbol Z³ and Z⁴ while denoting the symbol Z⁴ at least once; Z³ denotes a divalent radical which is a straight-chain or branched-chain alkylene or hydroxyalkylene radical having up to about 7 carbon atoms in the main chain, and Z⁴ is a divalent radical which is a straight-chain or branched-chain alkylene radical which has up to about 7 carbon atoms in the main chain, is unsubstituted and substituted by one or more hydroxyl radicals and is interrupted by one or more nitrogen atoms, the nitrogen atom being substituted by an alkyl chain having from 1 to 4 carbon atoms, preferably 4 carbon atoms, which is optionally interrupted by an oxygen atom and optionally contains one or more hydroxyl groups; and
- 30 35

- iii) the alkylation products, with alkyl and benzyl halides of 1 to 6 carbon atoms, alkyl tosylates or mesylates, and the oxidation products, of the polymers of the formulae (I) and (II) indicated above under i) and ii).

5 (5) Polyamino-polyamides prepared by the polycondensation of an acid compound with a polyamine. The acid compound can be organic dicarboxylic acids, aliphatic monocarboxylic and dicarboxylic acids containing a double bond, esters of the abovementioned acids, preferably the esters with lower alkanols having from 1 to 6 carbon atoms, and mixtures thereof. The polyamine is a bis-
10 primary or mono- or bis-secondary polyalkylene-polyamine wherein up to 40 mol% of this polyamine can be a bis-primary amine, preferably ethylenediamine, or a bis-secondary amine, preferably piperazine, and up to 20 mol% can be hexamethylenediamine.

(6) The above mentioned polyamino-polyamides can be alkylated and/or crosslinked. The alkylation can be carried out with glycidol, ethylene oxide, propylene oxide or acrylamide. The crosslinking is carried out by means of a
15 crosslinking agent such as:

- i) epihalogenohydrins, diepoxides, dianhydrides, unsaturated anhydrides and bis-saturated derivatives, in proportions of 0.025 to 0.35 mol of crosslinking agent per amine group of the polyamino-polyamide;
 - 20 ii) bis-halogenohydrins, bis-azetidinium compounds, bishalogenoacyldiamines and bis-(alkyl halides);
 - iii) oligomers obtained by reacting a compound chosen from the group comprising bis-halogenohydrins, bis-azetidinium compounds, bis-halogenoacyl-diamines, bis-(alkyl halides), epihalogenohydrins, diepoxides and bis-unsaturated derivatives, with another compound
25 which is a difunctional compound which is reactive towards the compound; and
 - iv) the quaternisation product of a compound chosen from the compounds ii) and the oligomers iii) and containing one or more tertiary amine groups which can be totally or partially alkylated with an alkylating agent preferably chosen from methyl or ethyl chlorides, bromides, iodides, sulphates, mesylates and tosylates, benzyl chloride or bromide, ethylene oxide, propylene oxide and glycidol, the crosslinking
30 being carried out by means of 0.025 to 0.35 mol, in particular of 0.025 to 0.2 mol and more particularly of 0.025 to 0.1 mol, of crosslinking agent per amine group of the polyamino-polyamide.
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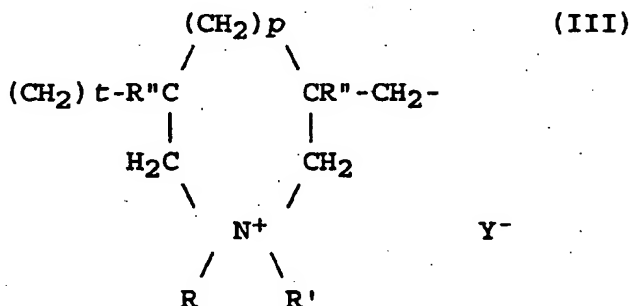
(7) Polyamino-polyamide derivatives resulting from the condensation of a polyalkylene-polyamine with a polycarboxylic acid, followed by alkylation by means of difunctional agents, such as the adipic acid/dialkylaminohydroxyalkyl-dialkylenetriamine copolymers in which the alkyl radical contains 1 to 4 carbon atoms and preferably denotes methyl, ethyl or propyl.

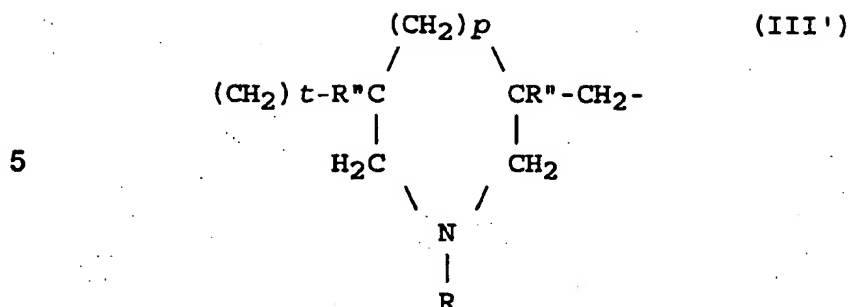
Useful polymers are adipic acid/dimethylaminohydroxypropyl-diethylenetriamine copolymers sold under the name Cartaretine F, F⁴ or F⁸ by SANDOZ.

(8) Polymers obtained by reacting polyalkylenepolyamine containing two primary amine groups and at least one secondary amine group, with a dicarboxylic acid chosen from diglycolic acid and saturated aliphatic dicarboxylic acids having 3 to 8 carbon atoms, the molar ratio of the polyalkylene-polyamine to the dicarboxylic acid being from 0.8:1 to 1.4:1, and the resulting polyamide being reacted with epichlorohydrin in a molar ratio of epichlorohydrin to the secondary amine groups of the polyamide of from 0.5:1 to 1.8:1.

Useful polymers are those sold under the name HERCOSETT 57 by Hercules Incorporated, and that sold under the name PD 170 or DELSETTE 101 by Hercules.

(9) Cyclic polymers generally having a molecular weight of 20,000 to 3,000,000 such as homopolymers containing, as the main constituent of the chain, units corresponding to the formula (III) or (III')

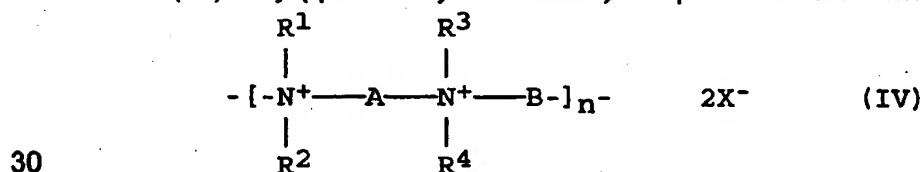




- 10 in which p and t are 0 or 1, and $p+t=1$, R'' denotes hydrogen or methyl, R and R' independently of one another denote an alkyl group having from 1 to 22 carbon-atoms, a hydroxylalkyl group in which the alkyl group preferably has 1 to 5 carbon atoms, or a lower amidoalkyl group, and R and R' can denote, together with the nitrogen atom to which they are attached, heterocyclic groups such as
- 15 piperidiny1 or morpholinyl, and Y is bromide, chloride, acetate, borate, citrate, tartrate, bisulphate, bisulphite, sulphate or phosphate. Copolymers containing units of the formula III and III' may also contain units derived from acrylamide or from diacetoneacrylamide.

20 Amongst the quaternary ammonium polymers of the type defined above, those which are preferred are the dimethyldiallylammonium chloride homopolymer sold under the name MERQUAT 100 and having a molecular weight of less than 100,000, and the dimethyldiallylammonium chloride/acrylamide copolymer having a molecular weight of more than 500,000 and sold under the name MERQUAT 550 by CALGON Corporation.

25 (10) Poly-(quaternary ammonium) compounds of the formula



wherein R^1 , R^2 , R^3 , and R^4 are independently aliphatic, alicyclic or arylaliphatic radicals containing a maximum of 20 carbon atoms, or lower hydroxylaliphatic radicals, or alternatively, with the nitrogen atoms to which they are attached, heterocyclic rings optionally containing a second hetero-atom other than

35 nitrogen, or alternatively R^1 , R^2 , R^3 , and R^4 represent a group $\text{CH}_2\text{CHR}^3\text{R}^4$ wherein R^3 denoting hydrogen or lower alkyl and R^4 denoting SO , CN , $\text{CON}(\text{R}^6)_2$, COOR^5 , COR^5 , COOR^7D , or CONHR^7D ; R^5 denoting lower alkyl, R^6 denoting hydrogen or lower alkyl, R^7 denoting alkylene and D denoting a quaternary ammonium group; A and B independently represent a polymethylene

group containing from 2 to 20 carbon atoms, which can be linear or branched, saturated or unsaturated and can contain, inserted in the main chain one or more groups $-\text{CH}_2\text{-Y-CH}_2-$ wherein Y denotes benzene, oxygen, sulfur, SO, SO_2 , SS, NR^8 , $\text{N}^+(\text{R}^9)_2\text{X}^{1-}$, CHOH , NHCONH , CONR^8 , or COO ; X^{1-} denoting an anion derived from a mineral or organic acid, R^8 denoting hydrogen or lower alkyl and R^9 denoting lower alkyl, or alternatively A and R^1 and R^3 form a piperazine ring with the two nitrogen atoms to which they are attached. If A denotes a linear or branched, saturated or unsaturated alkylene or hydroxyalkylene radical, B can also denote a group: $-(\text{CH}_2)_n\text{-CO-D-OC-(CH}_2)_n-$; wherein n is selected so that the molecular weight is generally between 1,000 and 100,000; and D denotes:

i) a glycol radical of the formula $-\text{O-Z-O-}$, in which Z denotes a linear or branched hydrocarbon radical or a group corresponding to the formulae:

$-\text{[CH}_2\text{-CH}_2\text{-O]}_x\text{-CH}_2\text{-CH}_2-$ or $-\text{[CH}_2\text{-C(CH}_3\text{)H-O]}_y\text{-CH}_2\text{-C(CH}_3\text{)H-}$ wherein x and y denote an integer from 1 to 4, representing a definite and unique degree of polymerisation;

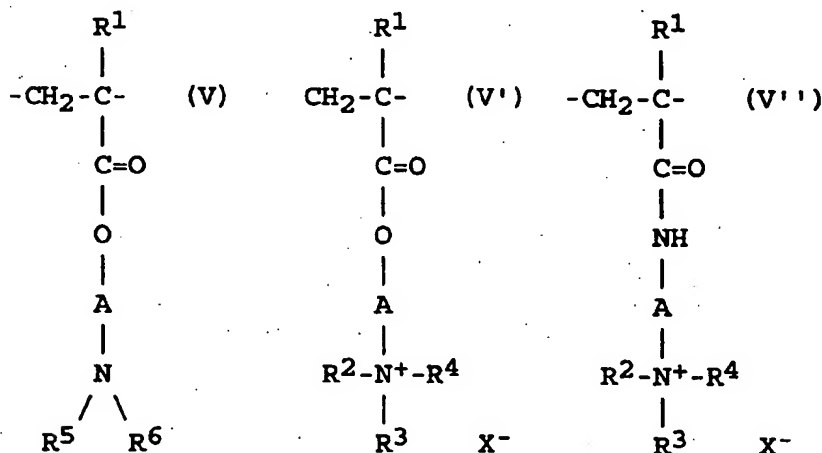
ii) a bis-secondary-diamine radical, such as a piperazine derivative;

iii) a bis-primary diamine radical of the formula: $-\text{N-H-Y-NH-}$, in which Y denotes a linear or branched hydrocarbon radical or the divalent radical

$-\text{CH}_2\text{-CH}_2\text{-S-S-CH}_2\text{-CH}_2-$; or

iv) a ureylene group of the formula $-\text{N-H-CO-NH-}$.

(11) Homopolymers or copolymers derived from acrylic or methacrylic acid and containing at least one unit:



wherein R¹ is H or CH₃, A is a linear or branched alkyl group having 1 to 6 carbon atoms or a hydroxyalkyl group having 1 to 4 carbon atoms, R², R³ and R⁴ independently denote an alkyl group having 1 to 18 carbon atoms or a benzyl group, R⁵ and R⁶ denote H or alkyl having 1 to 6 carbon atoms and X denotes methosulphate or halide, such as chloride or bromide.

The comonomer or comonomers which can be used typically belong to the family comprising: acrylamide, methacrylamide, diacetone-acrylamide, acrylamide and methacrylamide substituted on the nitrogen by one or more lower alkyls, alkyl esters of acrylic and methacrylic acids, vinylpyrrolidone and vinyl esters.

Useful polymers are Quaternium 38, 37, 49 and 42 in the CTFA, acrylamide/beta-methacryloyloxyethyl-trime-thylammonium methosulphate copolymers sold under the names Teten 205,210,220 and 240 by Hercules, and aminoethylacrylate phosphate/acrylate copolymer sold under the name Catrex by National Starch & Chemicals, and the crosslinked graft cationic copolymers having a molecular weight of 10,000 to 1,000,000, and preferably of 15,000 to 500,000, and resulting from the copolymerisation of: at least one cosmetic monomer, dimethylaminoethyl methacrylate, polyethylene glycol and a polyunsaturated crosslinking agent, such as those mentioned in the CTFA dictionary under the name AMODIMETHICONE, such as the product marketed as a mixture with other ingredients under the name DOW CORNING 929 cationic emulsion.

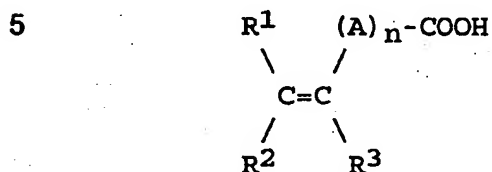
(12) Other cationic polymers which can be used are polyalkyleneimines, in particular polyethyleneimines, polymers containing vinylpyridine units or vinylpyridinium units in the chain, condensates of polyamines and of epichlorohydrin, poly-(quaternary ureylenes) and chitin derivatives.

Highly preferred cationic fixative polymers include commercially available material such as Polyquaternium 4 under the tradenames CELQUAT H100 and CELQUAT L200 supplied by National Starch & Chemicals, and Polyquaternium 11 under the tradename GAFQUAT 755N supplied by ISP.

Anionic Fixative Polymer

The anionic fixative polymers useful herein include polymers containing units derived from carboxylic, sulphonic or phosphoric acid and usually have a molecular weight of 500 to 5,000,000. These polymers are water-soluble polymers, it being possible for this solubility to be obtained by neutralisation.

The carboxylic acid groups can be provided by unsaturated monocarboxylic or dicarboxylic acids, such as those corresponding to the formula:



wherein n is 0 or an integer from 1 to 10, A denotes a methylene group optionally joined to the carbon atom of the saturated group, or to the adjacent methylene group in the case where n is greater than 1, via a heteroatom, such as oxygen or sulphur, R¹ denotes a hydrogen atom or a phenyl or benzyl group, R² denotes a hydrogen atom, a lower alkyl group or a carboxyl group and R³ denotes a hydrogen atom, a lower alkyl group, CH₂COOH, or a phenyl or benzyl group.

According to the invention, the preferred polymers containing carboxylic acid groups are:

(1) Homopolymers or copolymers of acrylic or methacrylic acid or salts thereof, and in particular, the products sold under the name VERSICOL E or K, and ULTRAHOLD by BASF and under the name DARVAN No. 7 by Van der Bilt; acrylic acid/acrylamide copolymers sold in the form of their sodium salt under the name RETEN 421, 423 or 425 by HERCULES; and the sodium salts of polyhydroxycarboxylic acids, sold under the name HYDAGEN F by HENKEL.

(2) Copolymers of acrylic or methacrylic acid with a monoethylenic monomer, such as ethylene, styrene, a vinyl or allyl ester or acrylic or methacrylic acid ester, optionally grafted onto a polyalkylene glycol, such as polyethylene glycol, and optionally crosslinked. Other such copolymers contain an optionally N-alkylated and/or N-hydroxylated acrylamide unit in their chain, such as those sold under the name QUADRAMER 5 by American Cyanamid.

(3) Copolymers derived from crotonic acid, such as those containing, in their chain, vinyl acetate or propionate units and optionally other monomers such as allyl or methallyl esters, a vinyl ether or a vinyl ester of a saturated linear or branched carboxylic acid with a hydrocarbon chain of at least 5 carbon atoms, if appropriate, for these polymers to be grafted and crosslinked, or also a vinyl, allyl or methallyl ester of an α- or β-cyclic carboxylic acid. Included in this class are

those with the tradename RESYN 28-2930, 28-2913, and 28-1310 sold by National Starch & Chemicals.

(4) Polymers derived from maleic, fumaric and itaconic acids or anhydrides with vinyl esters, vinyl ethers, vinyl halides, phenylvinyl derivatives, acrylic acid and its esters, such as those sold under the name GANTREZ A, SP, and ES by ISP. Other polymers included in this class are copolymers of maleic, citraconic and itaconic anhydrides with an allyl or methallyl ester optionally containing an acrylamido or methacrylamido group, or with an α -olefine, acrylic or methacrylic acid ester, acrylic or methacrylic acid or vinylpyrrolidone unit in their chain; the anhydride groups can be monoesterified or monoamidified.

(5) Polyacrylamides containing carboxylate groups. Polymers comprising sulphonic groups include polymers containing vinylsulphonic, styrenesulphonic, lignosulphonic or naphthalenesulphonic units. These polymers are chosen, in particular, from amongst:

- i) Polyvinylsulphonic acid salts having a molecular weight of 1,000 to 100,000, and also copolymers with an unsaturated comonomer, such as acrylic or methacrylic acid or an ester thereof and also substituted or unsubstituted acrylamide or methacrylamide, vinyl esters, vinyl ethers and vinylpyrrolidone.
- ii) Polystyrenesulphonic acid salts, such as the sodium salt sold by National Starch & Chemicals under the name Flexan 500 and 130.
- iii) Alkali metal or alkaline earth metal salts of sulphonic acids derived from lignin, and more particularly calcium lignosulphonates or sodium lignosulphonates, such as the product sold under the name Marasperse C-21 by American Can Co. and the C₁₀ to C₁₄ products sold by Avebene.
- iv) Polymers containing salified alkyl naphthalenesulphonic acid units, such as the sodium salt under the name Darvan No. 1 by Van der Bilt.

The anionic hair fixative polymers herein which include anionic monomers are preferably utilised in at least partially neutralised form in order to aid shampoo removability of the liquid hair cosmetic compositions. In the compositions the neutralisation of a polymer may be achieved by use of an inorganic base, preferably KOH. However organic base, preferably AMP (amino methyl propanol) and mixture of inorganic and organic base may also be used to effect the desired level of neutralisation in hair styling compositions. In total from about 50% to about 100%, preferably from about 70% to about 100%, most

preferably from about 80% to about 100% of the acidic monomers of each polymer utilised should be neutralised with base.

Any conventionally used base, organic or inorganic, may be used for neutralisation of acidic polymers provided they are utilised as specified herein.

- 5 Hydroxides of alkali, alkaline earth and amino alcohols are suitable neutralisers.

Examples of suitable organic neutralizing agents which may be included in the compositions of the present invention include amines, especially amino alcohols such as 2-amino-2-methyl-1, 3-propanediol (AMPD), 2-amine-2ethyl-1, 3-propanediol (AEPD), 2-amino-2-methyl-1-propanol (AMP), 2-amino-1-butanol (AB), monethanolamine (MEA), diethanolamine (DEA), triethanolamine (TEA),
10 monoisopropanolamine (MIPA), diisopropanolamine (DIPA), triisopropanolamine (TIPA), dimethylsteramine (DMS) and amino methyl propanol (AMP) and mixtures thereof.

- 15 Preferred neutralising agents for use in hair care compositions of the present invention are potassium and sodium hydroxides.

Highly preferred anionic fixative polymers include commercially available material such as vinyl acetate/crotonic acid/vinyl neodecanoate copolymers and vinyl acetate/crotonic acid copolymers with the tradenames RESYN 28-2930, RESYN 28-2913, and RESYN 28-1310 supplied by National Starch & Chemicals,
20 and acrylates copolymers and acrylates/acrylamide copolymers with tradenames LUVIMER 100P, ULTRAHOLD 8, and ULTRAHOLD STRONG supplied by BASF Corporation.

Nonionic Fixative Polymers

- 25 Nonionic fixative polymers useful herein are homopolymer of vinylpyrrolidone or vinylcaprolactum and copolymers of vinylpyrrolidone with vinylacetate such as those with tradenames LUVISKOL K grades and LUVISKOL VA grades supplied by BASF Corporation.

Silicone Grafted Copolymers

- 30 The silicone grafted copolymers useful herein include those which have a vinyl polymeric backbone (A and B monomers), and grafted to such backbone a polydimethylsiloxane macromer (C macromer) having a weight average molecular weight of from about 1,000 to about 50,000. Preferably, these copolymers contain from and from about 50.0% to about 99.9% of the combination of A and B monomers, and from about 0.1% to about 50.0% of C
35 macromer, wherein A is a lipophilic, low polarity free radically polymerizable vinyl monomer, such as methacrylic or acrylic esters; B is a hydrophilic polar monomer

which is copolymerizable with A, such as acrylic acid, N,N-dimethylacrylamide, dimethylaminoethylmethacrylate, diallyldimethylammonium chloride, vinyl pyrrolidone, or quaternized dimethylaminoethyl methacrylate; and C is a silicone-containing macromer having a weight average molecular weight of from about 1,000 to about 50,000, based on polydimethylsiloxane. Preferred silicone grafted copolymers are those having a T_g above about -20°C, and a molecular weight of from about 10,000 to about 1,000,000.

Suitable silicone grafted copolymers herein include those listed below wherein the numbers indicate the weight ratio of monomers and macromers in the copolymer, silicone macromer S1 is a dimethylpolysiloxane having a molecular weight of about 20,000, and silicone macromer S2 is a dimethylpolysiloxane having a molecular weight of about 10,000.

- 1) 20/63/17 acrylic acid/t-butyl methacrylate/silicone macromer S2, having a copolymer molecular weight about 100,000
- 2) 10/70/20 dimethylacrylamide/isobutyl methacrylate/silicone macromer S2, having a copolymer molecular weight about 400,000
- 3) 60/20/20 diallyldimethyl ammonium methacrylate/isobutyl methoacrylate /silicone macromer S2, having a copolymer molecular weight about 500,000
- 4) 40/40/20 acrylic acid/methyl methacrylate/silicone macromer S1, having a copolymer molecular weight about 400,000
- 5) 10/70/20 acrylic acid/t-butyl methacrylate/silicone macromer S1, having a copolymer molecular weight about 300,000
- 6) 25/65/10 acrylic acid/isopropyl methacrylate/silicone macromer S2, having a copolymer molecular weight about 200,000
- 7) 60/25/15 N,N'-dimethylacrylamide/methoxyethyl methacrylate/silicone macromer S1, having a copolymer molecular weight about 200,000
- 8) 12/64/4/20 N,N'-dimethylacrylamide/isobutyl methacrylate/2-ethylhexyl methacrylate/silicone macromer S1, having a copolymer molecular weight about 300,000
- 9) 30/40/10/20 N,N'-dimethylacrylamide/isobutyl methacrylate/2-ethylhexyl methacrylate/silicone macromer S1, having a copolymer molecular weight about 300,000
- 10) 80/20 t-butylacrylate/silicone macromer S2, having a copolymer molecular weight about 150,000

CARRIER

Composition herein may include a carrier. The level and species of the carrier are selected according to the compatibility with other components, and desired characteristic of the product. For example, a high percentage of volatile solvents of low boiling point and/or propellant are suitably used for product forms aimed to be left on the hair. On the other hand, water solutions of volatile and non-volatile solvents are suitably used for product forms aimed to be rinsed off the hair after washing or treating the hair with the product.

The carrier useful in the present invention include volatile solvents, non-volatile solvents, propellants, and mixtures thereof.

10 Volatile solvents useful herein include water, lower alkyl alcohols having from 1 to 3 carbons, and hydrocarbons having from about 5 to about 8 carbons. The preferred volatile solvents are water, ethanol, isopropanol, pentane, hexane, and heptane. The water useful herein include deionized water and water from natural sources containing mineral cations. Deionized water is preferred.

15 Non-volatile solvents useful herein include alkyl alcohols having more than 3 carbons, and polyhydric alcohols. The polyhydric alcohols useful herein include 1,2-propane diol or propylene glycol, 1,3-propane diol, hexylene glycol, glycerin, diethylene glycol, dipropylene glycol, 1,2-butylene glycol, and 1,4-butylene glycol.

20 Propellants may be used for mousse and hair spray product forms. Propellants, when used in the present invention, are selected depending on variables such as the remainder of components, the package, and whether the product is designed to be used standing or invert.

25 Propellants useful herein include fluorohydrocarbons such as difluoroethane 152a available from DuPont, dimethylether, and hydrocarbons such as propane, isobutane, n-butane, mixture of hydrocarbons such as LPG (liquid petroleum gas), carbon dioxide, nitrous oxide, nitrogen, and compressed air.

ADDITIONAL COMPONENTS

30 The compositions of the present invention may include a variety of additional components, which may be selected by the artisan according to the desired characteristics of the final product. Additional components include, for example, polyvalent metal cations, suspending agents, and other additional components.

35 Polyvalent Metal Cations

Suitable polyvalent metal cations include divalent and trivalent metals, divalent metals being preferred. Exemplary metal cations include alkaline earth metals, such as magnesium, calcium, zinc, and copper, and trivalent metals such as aluminum and iron. Preferred are calcium and magnesium.

5 The polyvalent metal cation can be added as an inorganic salt, organic salt, or as a hydroxide. The polyvalent metal cation may also be added as a salt with anionic surfactants as mentioned above.

Preferably, the polyvalent metal cation is introduced as an inorganic salt or organic salt. Inorganic salts include chloride, bromide, iodine, nitrate, or sulfate,
10 more preferably chloride or sulfate. Organic salts include L-glutamate, lactate, malate, succinate, acetate, fumarate, L-glutamic acid hydrochloride, and tartarate.

It will be clear to those skilled in the art that, if polyvalent salts of the anionic surfactant is used as the mode of introducing the polyvalent metal cations
15 into the compositions hereof, only a fraction of the anionic surfactant may be of polyvalent form, the remainder of the anionic surfactant being necessarily added in monovalent form.

Hardness of the conditioning shampoo compositions can be measured by standard methods in the art, such as by ethylene diamine tetraacetic acid (EDTA)
20 titration. In the event that the composition contains dyes or other color materials that interfere with the ability of EDTA titration to yield a perceptible color change, hardness should be determined from the composition in the absence of the interfering dye or color.

Suspending Agents

25 A preferred additional component is a suspending agent, particularly for compositions comprising silicone compounds of high viscosity and/or large particle size. When present, the suspending agent is in dispersed form in the compositions. The suspending agent will generally comprise from about 0.1% to about 10%, and more typically from about 0.3% to about 5.0%, by weight, of the
30 composition.

Preferred suspending agents include acyl derivatives such as ethylene glycol stearates, both mono and distearate, long chain amine oxides such as alkyl (C₁₆-C₂₂) dimethyl amine oxides, e.g., stearyl dimethyl amine oxide, and mixtures thereof. When used in the shampoo compositions, these preferred
35 suspending agents are present in the composition in crystalline form. These suspending agents are described in U.S. Patent 4,741,855.

Other suitable suspending agents include alkanol amides of fatty acids, preferably having from about 16 to about 22 carbon atoms, more preferably about 16 to 18 carbon atoms, preferred examples of which include stearic monoethanolamide, cocomonethanolamide, stearic diethanolamide, stearic monoisopropanolamide and stearic monoethanolamide stearate.

Other suitable suspending agents include N,N-dihydrocarbyl amido benzoic acid and soluble salts thereof (e.g., Na and K salts), particularly N,N-di(hydrogenated) C₁₆, C₁₈ and tallow amido benzoic acid species of this family, which are commercially available from Stepan Company (Northfield, Illinois, USA).

Other suitable suspending agents include xanthan gum. The use of xanthan gum as a suspending agent in silicone containing shampoo compositions is described, for example, in U.S. Patent 4,788,006, which is incorporated herein by reference in its entirety. Combinations of long chain acyl derivatives and xanthan gum may also be used as a suspending agent in the shampoo compositions. Such combinations are described in U.S. Patent 4,704,272, which is incorporated herein by reference in its entirety.

Other suitable suspending agents include carboxyvinyl polymers. Preferred among these polymers are the copolymers of acrylic acid crosslinked with polyallylsucrose as described in U.S. Patent 2,798,053, which is incorporated herein by reference in its entirety. Examples of these polymers include the carbomers, which are homopolymers of acrylic acid crosslinked with an allyl ether of pentaerythritol, an allyl ether of sucrose, or an allyl ether of propylene. Neutralizers may be required, for example, amino methyl propanol, triethanol amine, or sodium hydroxide.

Other suitable suspending agents can be used in the compositions, including those that can impart a gel-like viscosity to the composition, such as water soluble or colloidally water soluble polymers like cellulose ethers such as hydroxyethyl cellulose, hydroxymethyl cellulose, hydroxypropyl cellulose, and materials such as guar gum, polyvinyl alcohol, polyvinyl pyrrolidone, hydroxypropyl guar gum, starch and starch derivatives.

Other Additional Components

A wide variety of other additional ingredients can be formulated into the present compositions. These include: other conditioning agents such as hydrolysed collagen with tradename Peptin 2000 available from Hormel, vitamin E with tradename Emix-d available from Eisai, panthenol available from Roche,

panthenyl ethyl ether available from Roche, hydrolysed keratin, proteins, plant extracts, and nutrients; emulsifying surfactants for dispersing water insoluble components in the carrier; preservatives such as benzyl alcohol, methyl paraben, propyl paraben and imidazolidinyl urea; pH adjusting agents, such as citric acid, sodium citrate, succinic acid, phosphoric acid, sodium hydroxide, sodium carbonate; salts, in general, such as potassium acetate and sodium chloride; coloring agents, such as any of the FD&C or D&C dyes; hair oxidizing (bleaching) agents, such as hydrogen peroxide, perborate and persulfate salts; hair reducing agents such as the thioglycolates; perfumes; and sequestering agents, such as disodium ethylenediamine tetra-acetate; ultraviolet and infrared screening and absorbing agents such as octyl salicylate, and antidandruff agents such as zinc pyridinethione. Such optional ingredients generally are used individually at levels from about 0.001% to about 10.0%, preferably from about 0.01% to about 5.0% by weight of the composition.

EXAMPLES

The following examples further describe and demonstrate embodiments within the scope of the present invention. The examples are given solely for the purpose of illustration and are not to be construed as limitations of the present invention, as many variations thereof are possible without departing from the spirit and scope of the invention. Ingredients are identified by chemical or CTFA name, or otherwise defined below.

Examples I - V Leave on sprays.

	Ex.I	Ex.II	Ex.III	Ex.IV	Ex.V
Components	Percentage by weight				
1,4-bis(2-cyanostyryl)benzene*1	0.50	-	-	-	-
Disodium-1,4-bis(2-sulfostyryl)biphenyl*2	-	0.50	-	-	-
4,4'-bis[(4-anilino-6-bis(2-hydroxyethyl)amino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disulfonic acid*3	-	-	1.00	-	-
4-methyl-6,7-dihydroxycoumarin	-	-	-	0.25	-
4-methyl-7,7'-dimethylamino coumarin	-	-	-	0.25	-

4,4'-bis(5-methylbenzoxazol-2-yl)stilbene*4	-	-	-	-	0.50
Carbomer 956*5	0.26	0.26	-	-	-
Citric Acid*6	0.10	0.10	-	-	-
L-Glutamic Acid*7	-	-	0.10	0.10	0.10
Octyl Methoxycinnamate*8	0.01	0.01	-	-	-
Stearamidopropyl dimethylamine*9	-	-	0.50	0.50	0.50
Silicone Emulsion*10	5.80	5.80	-	-	-
Preservatives	0.65	0.65	0.65	0.65	0.65
Perfume	0.15	0.15	0.08	0.08	0.08
Hydrolyzed Collagen*37	0.01	-	-	-	-
Vitamin E*38	0.01	-	-	-	-
Panthenol*39	0.025	-	-	-	-
Panthenyl Ethyl Ether*40	0.025	-	-	-	-
Deionized water	----- q.s. to 100 -----				
Color Alteration	Average Values				
a value					
treated side	0.63	2.51	2.90	2.27	0.44
control side	0.81	2.05	2.79	2.88	1.16
alteration	0.18	0.46	0.11	0.61	0.72
b value					
treated side	0.11	1.11	1.61	1.70	0.77
control side	0.58	2.60	2.98	2.67	1.16
alteration	0.47	1.49	1.37	0.97	0.39

Examples VI- IX Leave on hair lotions.

	Ex.VI	Ex.VII	Ex.VIII	Ex.IX
Components	Percentage by weight			
Disodium-1,4-bis(2-sulfostyryl)biphenyl*2	1.00	0.20	-	-
2-(4-styryl-3-sulfophenyl)-2H-naphtho[1,2-d]triazole*11	-	0.20	-	-
1-(4-amidosulfonylphenyl)-3-(4-chlorophenyl)-2-pyrazoline*12	-	-	1.00	-
2,4-dimethoxy-6-(1'pyrenyl)-1,3,5-triazine*13	-	-	-	0.50
Acrylates/Steareth-20 Methacrylate Copolymer*14	1.75	-	-	-
Aminomethylpropanol*15	2.10	-	-	-
Cetyl Alcohol*16	2.00	0.50	0.50	0.50
Cetyl Hydroxyethyl Cellulose*17	-	0.75	0.75	0.75
Citric Acid*6	0-0.2	0-0.2	0-0.2	0-0.2
Lauryl Methyl Gluceth-10*18 Hydroxypropyldimonium Chloride*19	-	1.00	1.00	1.00
Pentaerythritol Tetraisostearate*20	1.00	3.00	3.00	3.00
Polyoxyethylene Glycol*21	0.50	0.50	0.50	0.50
Polyquaternium-10*22	0.10	-	-	-
Polyquaternium-7*23	-	0.50	0.50	0.50
Polysorbate 60*24	1.00	-	-	-
Silicone Emulsion*25	1.00	0.50	0.50	0.50
Stearyl Alcohol*26	3.00	0.50	0.50	0.50
Preservatives	0.90	0.90	0.90	0.90
Perfume	0.08	0.08	0.08	0.08
Deionized water	q.s. to 100			
Color Measurement	Average Values			
a value				
treated side	1.20	1.57	2.55	0.18

control side	1.80	1.64	2.68	0.74
alteration.	0.60	0.07	0.13	0.56
b value				
treated side	0.20	0.12	1.31	0.45
control side	1.80	1.28	2.89	0.55
alteration	2.00	1.40	1.58	1.00

Examples X - XIII Shampoos and Conditioners.

	Ex.X	Ex.XI	Ex.XII	Ex.XIII
Components	Percentage by weight			
Disodium-1,4-bis(2-sulfoxyethyl)biphenyl*2	0.50	-	0.50	1.00
4,4'-bis[(4-anilino-6-bis(2-hydroxyethyl)amino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disulfonic acid*3	-	1.00	-	-
Cetyl Alcohol*16	-	-	2.50	2.50
L-Glutamic Acid*7	-	-	0.64	0.64
Polyoxyethylene stearyl ether*27	-	-	-	0.50
Polyquaternium-10*22	0.50	0.50	-	-
Stearamidopropyl dimethylamine*9	-	-	2.00	2.00
Stearyl Alcohol*26	-	-	4.50	4.50
Preservatives	-	-	0.53	0.53
Perfume	0.50	0.50	0.20	0.20
Ammonium Laureth-3 Sulfate	10.00	10.00	-	-
N-acyl-L-glutamate*28	4.00	4.00	-	-
Triethanolamine				
Cocamidopropylbetaine*29	4.00	4.00	-	-
Cocamide MEA	1.50	1.50	-	-
Ethylene Glycol Distearate*30	2.00	2.00	-	-
DMDM Hydantoin	0.20	0.20	-	-
15/85 Silicone Blend*31	-	-	4.20	4.20
40/60 Silicone blend*32	2.00	2.00	-	-

Monosodium phosphate	0-1.0	0-1.0	-	-
Disodium phosphate	0-1.0	0-1.0	-	-
Deionized water	q.s. to 100			
Color Measurement	Average Values			
a value				
treated side	1.80	0.37	0.76	0.85
control side	2.03	0.59	1.10	0.92
alteration	0.23	0.22	0.34	0.07
b value				
treated side	0.83	0.13	0.04	0.12
control side	1.66	0.47	0.64	0.63
alteration	0.83	0.60	0.60	0.51

Examples XIV- XVI Fixative hair spray and mousses:

	Ex.XIV	Ex.XV	Ex.XVI
Components	Percentage by weight		
Disodium-1,4-bis(2-sulfostyryl)biphenyl*2	0.20	0.25	-
porphyrin*33	-	-	0.20
Denatured Ethyl Alcohol	73.25	7.50	7.50
Perfume	0.20	0.10	0.10
Polyacrylic acid-co-ethyl acrylate-co-t-butyl acrylamide*34	6.00	-	-
Polyoctylacrylamide-co-alkylacrylate-co-butylaminoethyl methacrylate*35	-	4.00	4.00
Aminomethyl propanol*15	0.15	0.70	0.70
Dimethicone copolyol*36	0.10	-	-
Lauramide DEA	0.10	-	-
Laureth-23	-	0.20	0.20
Laureth-7	-	0.10	0.10
Propylene glycol	-	0.10	0.10
Preservatives	-	0.10	0.10
isobutane/propane	-	7.00	7.00
Deionized water	q.s. to 100		

Color Measurement	Average Values		
a value			
treated side	1.31	1.30	1.10
control side	1.66	1.14	1.25
alteration	0.35	0.16	0.15
b value			
treated side	4.25	0.38	0.32
control side	5.30	1.16	1.23
alteration	1.05	0.78	0.91

Definitions

- *1 1,4-bis(2-cyanostyryl)benzene: Ultraphor RN obtained by BASF
- *2 Disodium-1,4-bis(2-sulfostyryl)biphenyl: Tinopal CBX obtained by Ciba Geigy
- 5 *3 4,4'-bis[(4-anilino-6-bis(2-hydroxyethyl)amino-1,3,5-triazin-2-yl)amino]stilbene-2,2'-disulfonic acids: Tinopal UNPA-GX obtained by Ciba Geigy
- *4 4,4'-bis(5-methylbenzoxazol-2-yl)stilbene: available from TCI
- *5 Carbomer 956: Carbopol 980 obtained by BF Goodrich
- 10 *6 Citric Acid: Anhydrous Citric acid obtained by Haarmann & Reimer
- *7 L-Glutamic Acid: L-Glutamic obtained acid (cosmetic grade) by Ajinomoto
- *8 Octyl Methoxycinnamate: available from ISP
- *9 Stearylamidopropyl dimethylamine: Amidoamine MPS obtained by Nikko
- *10 Silicone Emulsion: SM-2169 obtained by G.E.
- 15 *11 2-(4-styryl-3-sulfophenyl)-2H-naphto[1,2-d]triazole: Tinopal RBS available from Ciba Specialty Chemicals
- *12 1-(4-amidosulfonylphenyl)-3-(4-chlorophenyl)-2-pyrazoline: Blankophor DCB obtained by Bayer
- *13 2,4-dimethoxy-6-(1'pyrenyl)-1,3,5-triazine: available from Ciba Geigy
- 20 *14 Acrylates/Steareth-20 Methacrylate Copolymer: Aculyn obtained by Rohm&Haas
- *15 Aminomethylpropanol: AMP-regular obtained by Angus
- *16 Cetyl Alcohol : Konol series obtained by Shinihon Rika
- *17 Cetyl Hydroxyethyl Cellulose: Polysurf 67 obtained by Aqualon
- 25 *18 Lauryl Methyl Gluceth-10 Hydroxypropyldimonium Chloride: Glucquat 125 obtained by Amerchol
- *20 Pentaerythritol Tetraisostearate: KAK PTI obtained by Kokyu alcohol

- *21 Polyoxyethylene Glycol: WSR N-10 obtained by Amerchol
- *22 Polyquaternium-10: UCARE Polymer LR 400 obtained by Amerchol
- *23 Polyquaternium-7: MerquatS obtained by Calgon
- *24 Polysorbate 60: Tween 60 obtained by ICI
- 5 *25 Silicone Emulsion: X65-4829 obtained by Tosil/GE
- *26 Stearyl Alcohol: Konol series obtained by Shinihon Rika
- *27 Polyoxyethylene stearyl ether: Brij-721 obtained by ICI
- *28 N-acyl-L-glutamate Triethanolamine: CT12S obtained by Ajinomoto
- *29 Cocamidopropylbetaine: Tegobetain obtained by Th.Goldschmidt AG
- 10 *30 Ethylene Glycol Distearate: EGDS obtained by Th. Goldschmidt AG
- *31 15/85 Silicone Blend: available from Shinetsu
- *32 40/60 Silicone blend: SE76 obtained by G.E.
- *33 porphyrin: available from Wako
- *34 Polyacrylic acid-co-ethyl acrylate-co-t-butyl acrylamide: Ultrahold-8
- 15 obtained by BASF
- *35 Polyoctylacrylamide-co-alkylacrylate-co-butylaminoethyl methacrylate: Amphomer 28-4910 obtained by National Starch
- *36 Dimethicone copolyol: SH3746 obtained by Dow Corning
- *37 Hydrolyzed collagen: Peptin 2000 obtained by Hormel
- 20 *38 Vitamin E: Emix-d obtained by Eisai
- *39 Panthenol: available from Roche
- *40 Panthenyl Ethyl Ether: available from Roche

Method of Preparation

25 The compositions of Examples I through XVI as shown above can be prepared by any conventional method well known in the art. Suitable methods are described below.

The hair spray of Examples I and II are suitably made as follows: Water, Carbomer 956, Optical brightener Ultraphor RN or Tinopal CBS-X are mixed at room temperature. Dimethylpolysiloxane emulsion SM2169 is added. A
30 triblender and mill is used to properly disperse the materials. Following this the remaining components are added with agitation.

The hair spray of Examples III, IV, and V are suitably made as follows: Water, Stearamidopropyl dimethylamine, Glutamic acid, Cetyl, and Stearyl, alcohol is heated above 70°C and then gradually cooled down to room
35 temperature. Silicone emulsion is added. A triblender and mill is used to properly

disperse the materials. Following this the remaining components are added with agitation.

The hair lotion of example VI is suitably made as follows: Acrylates/Steareth-20 Methacrylate Copolymer is neutralized with
5 Aminomethylpropanol in water at room temperature. Following this Polyquaternium-10 and Polyoxyethylene Glycol is added. The mixture is heated up to above 60°C and Cetyl alcohol, Stearyl alcohol and Polysorbate 60 is added. After cooling down to below 50°C the remaining components are added with agitation then cooled down to about 30°C.

10 The hair lotion of Examples VII, VIII, and IX are suitably made as follows : Disperse Cetyl Hydroxyethyl Cellulose and Polyquaternium-10 or Polyquaternium-7 in water at room temperature. Following this Polyoxyethylene Glycol is added. The mixture is heated up to above 60°C and Cetyl alcohol, Stearyl alcohol and Lauryl Methyl Gluceth-10 Hydroxypropyldimonium chloride is
15 added. After cooling down to below 50°C the remaining components are added with agitation then cooled down to about 30°C.

The shampoo of Examples X and XI are suitably made as follows: Water, Ammonium Laureth-3 Sulfate, Cetyl Alcohol, Stearyl alcohol, Cocoamide MEA, and Ethylene Glycol Distearate are heated above 75°C and then gradually
20 cooled down to room temperature. Polyquaternium-10 is separately dispersed in water at 35°C. The above two are mixed together and the remaining components are added with agitation.

The conditioner of Example XII is suitably made as follows: Water, Stearamidopropyldimethylamine, Glutamic acid, Cetyl, Stearyl alcohol is heated
25 above 70°C and then gradually cooled down to room temperature. Disodium-1,4'-bis(2-sulfostyryl)bisphenyl, Silicone blend is added. Following this the remaining components are added with agitation.

The conditioner of Example XIII is suitably made as follows: Water, Stearamidopropyldimethylamine, Glutamic acid, Disodium-1,4'-bis(2-sulfostyryl)bisphenyl is mixed and heated above 70°C. Polyethyleneoxide stearyl
30 ether is added. Separately, Water, Stearamidopropyldimethylamine, Glutamic acid, Cetyl, Stearyl alcohol is mixed and heated above 70°C. The above two mixtures are mixed together and cool to room temperature. The remaining components are added with agitation.

35 The fixative hair spray of Example XIV is suitably made as follows: Hair fixative polymer is dissolved in a portion of water and ethanol. To this is added

the remaining components except isobutane/propane blend. The obtained mixture is mixed until homogeneous. Finally, the concentrate thus obtained is packed into an aerosol can with isobutane/propane blend.

The hair mousse of Examples XV and XVI are suitably made as follows:

- 5 Hair fixative polymer is dissolved in a portion of water. To this is added the remaining ingredients except isobutane/propane blend. The obtained mixture is mixed until homogeneous. The concentrate thus obtained is packed into an aerosol can with isobutane/propane blend.

Method of Measuring Color Alteration

- 10 The color alteration as shown above is measured as described below.

- The hair color measurements are performed on hair switches having 20cm length made of 15g flat oriental hair. The same switch is separated into two parts, one part to be treated with the Example product (treated side), and the other part to be treated with a control product with the same formulation as the
- 15 Example product except for the optical brightener replaced by water(control side). To both sides of the hair switch, the same amount of product is applied as described in detail below. The measurement of the a-value and the b-value is done by either a Macbeth 1500 Reflective Color Meter (supplied by Kollmorgen Corporation) or on a MiniScan XE spectrophotometer (supplied by Hunter
- 20 Associates Laboratory Inc). Each measurement is repeated at least three times, and the average is obtained.

For each of Examples I-V, the Example product is sprayed on one part of the hair switch. The control product is sprayed on the other part. After drying, hair color is measured.

- 25 For each of Examples VI-IX, the Example product is applied on one part of the hair switch (pre-cleaned by ALS twice and dried overnight prior to use). The control product is applied on the other part of the hair switch. After drying, hair color is measured.

- For each of Examples X-XIII, the Example product is applied on one part
- 30 of the hair switch for 30 seconds, and rinsed by water for 30 seconds. The control product is applied on the other part of the hair switch for 30 seconds and rinsed by water for 30 seconds. After drying, hair color is measured.

- For Example XIV, the Example product is sprayed on one part of the 15 hair switch. The control product is sprayed on the other part. After drying, hair
- 35 color is measured.

For Examples XV and XVI, the Example product is applied on one part of the hair switch. The control product is applied on the other part of the hair switch. After drying, hair color is measured.

5 Examples I through XVI have many advantages. For example, they can be used daily, and provide color alteration, shininess, and UV protection to the hair.

10 It is understood that the examples and embodiments described herein are for illustrative purposes only and that various modifications or changes in light thereof will be suggested to one skilled in the art without departing from its spirit and scope.

WHAT IS CLAIMED IS:

1. A hair care composition comprising:
 - (a) an optical brightener in an amount which alters the color of the hair to which the composition is applied by from about 0.2 units to about 15 units of either or both of an a-value and a b-value; and
 - 5 (b) a hair care active selected from the group consisting of a deterative surfactant, a conditioning agent, a fixative polymer, and mixtures thereof.
2. The hair care composition according to Claim 1 further comprising a carrier.
3. The hair care composition according to Claim 1 wherein the absolute values of either or both of the a-value and the b-value are decreased.
4. The hair care composition according to Claim 1 wherein the absolute values of either or both of the a-value and the b-value are increased.
5. The hair care composition according to Claim 3 wherein the b-value is altered.
6. The hair care composition according to Claim 5 wherein the b-value is altered from about 0.5 units to about 5 units.
7. The hair care composition according to Claim 4 wherein the b-value is altered.
8. The hair care composition according to Claim 7 wherein the b-value is altered from about 0.5 units to about 5 units.
9. The hair care composition according to Claim 3 wherein the a-value is altered.
10. The hair care composition according to Claim 9 wherein the a-value is altered from about 0.5 units to about 5 units.

11. The hair care composition according to Claim 4 wherein the a-value is altered.

12. The hair care composition according to Claim 11 wherein the a-value is altered from about 0.5 units to about 5 units.

13. The hair care composition according to any of Claims 1 through 12 wherein the a-value is altered within about 5 units from the zero point of the a-value axis, and wherein the b-value is altered within about 5 units from the zero point of the b-value axis.

5

14. The hair care composition according to any of Claims 1 through 12 wherein the hair care composition is substantially free of dye-stuff.

15. The hair care composition according to any of Claims 1 through 12 wherein the optical brightener is selected from the group consisting of polystyrylstilbenes, triazinstilbenes, hydroxycoumarins, aminocoumarins, triazoles, pyrazolines, oxazoles, pyrenes, porphyrins, imidazoles, and mixtures thereof.

5

16. A hair care composition comprising:

(a) an optical brightener in an amount such that, when applied to the hair, is from about 1ppm to about 2% by weight of the hair; and

(b) a hair care active selected from the group consisting of a deterative surfactant, a conditioning agent, a fixative polymer, and mixtures thereof;

5

wherein the hair care composition alters the color of the hair to which the composition is applied by from about 0.2 units to about 15 units of either or both of an a-value and a b-value.

17. The hair care composition according to Claim 16 further comprising a carrier.

18. The hair care composition according to Claim 16 wherein the optical brightener is comprised in an amount such that, when applied to the hair, is from about 10ppm to about 1% by weight of the hair.

19. The hair care composition according to Claim 18 wherein the optical brightener is comprised in an amount such that, when applied to the hair, is from about 100ppm to about 0.3% by weight of the hair.

20. The hair care composition according to any of Claims 16 through 19 wherein the hair care composition is substantially free of dye-stuff.

21. The hair care composition according to any of Claims 16 through 19 wherein the optical brightener is selected from the group consisting of polystyrylstilbenes, triazinstilbenes, hydroxycoumarins, aminocoumarins, triazoles, pyrazolines, oxazoles, pyrenes, porphyrins, imidazoles, and mixtures thereof.

22. A method of altering the color of the hair by applying an optical brightener to the hair.

23. The method of altering the color of the hair according to Claim 22 wherein the optical brightener is applied to the hair in an amount from about 1ppm to about 2% by weight of the hair.

24. The method of altering the color of the hair according to Claim 22 wherein the optical brightener is selected from the group consisting of polystyrylstilbenes, triazinstilbenes, hydroxycoumarins, aminocoumarins, triazoles, pyrazolines, oxazoles, pyrenes, porphyrins, imidazoles, and mixtures thereof.

25. A method of altering the color of the hair by applying to the hair a composition comprising:

- (a) an optical brightener in an amount which alters the color of the hair to which the composition is applied by from about 0.2 units to about 15 units of either or both of an a-value and a b-value; and
- (b) a hair care active selected from the group consisting of a deterative surfactant, a conditioning agent, a fixative polymer, and mixtures thereof.

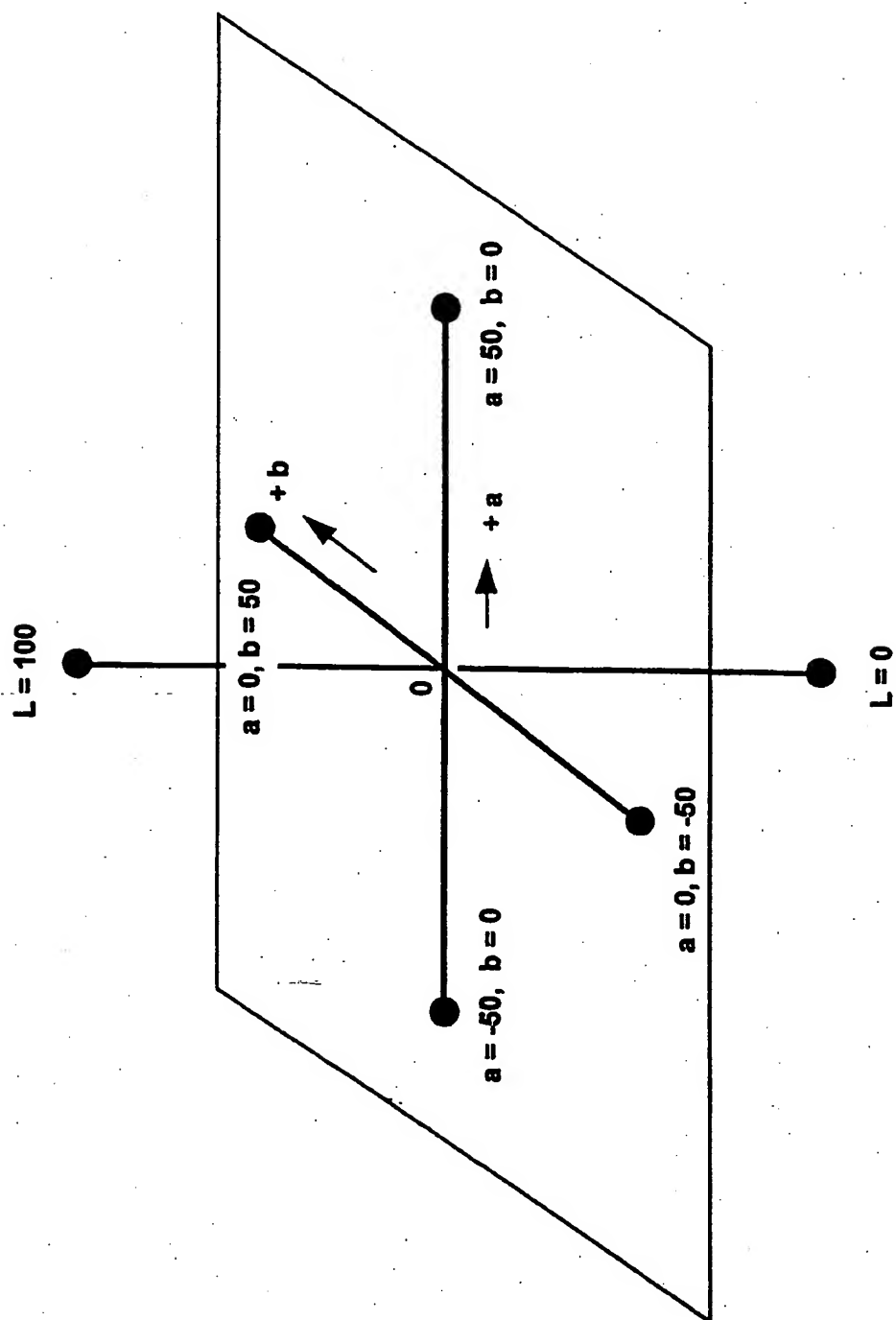


Fig. 1

INTERNATIONAL SEARCH REPORT

Int. Application No.
PCT/US 97/16410

A. CLASSIFICATION OF SUBJECT MATTER IPC 6 A61K7/13 A61K7/06		
According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) IPC 6 A61K		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched:		
Electronic data base consulted during the international search (name of data base and, where practical, search terms used)		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	EP 0 717 978 A (CURTIS HELENE IND INC) 26 June 1996 see claims 1,13,35; examples 35,72	1-25
X	GB 2 307 639 A (GEN ELECTRIC) 4 June 1997 cited in the application see page 8, line 3 - page 9, line 14; claims 1-9,11,12	1-25
X	US 3 658 985 A (OLSON FRANK WESLEY JR ET AL) 25 April 1972 cited in the application see column 1, line 62 - column 2, line 27; claims 1-13; examples 1-5	1-25
-/--		
<div style="display: flex; justify-content: space-between;"> <input checked="" type="checkbox"/> Further documents are listed in the continuation of box C. <input checked="" type="checkbox"/> Patent family members are listed in annex. </div>		
* Special categories of cited documents : <div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div> <div style="width: 45%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&" document member of the same patent family</p> </div> </div>		
Date of the actual completion of the international search <div style="text-align: center;">25 June 1998</div>		Date of mailing of the international search report <div style="text-align: center;">02/07/1998</div>
Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016		Authorized officer <div style="text-align: center;">Veronese, A</div>

INTERNATIONAL SEARCH REPORT

International Application No.

PCT/US 97/16410

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	GB 746 864 A ((COLGATE PALMOLIVE)) 21 March 1956 cited in the application see column 1, line 11-24 see column 2, line 59 - column 3, line 69 see claims; examples 1-3	1-25
X	US 4 312 855 A (GRAND PAUL S) 26 January 1982 cited in the application see column 5, line 45 - column 7, line 3 see claims; example 11 * Table III *	1-25
E	EP 0 834 303 A (KAO CORP) 8 April 1998 see examples 9,13	1-13, 15-25
X	US 3 577 528 A (EDMAN WALTER W ET AL) 4 May 1971 cited in the application see example 7	1-25
X	GB 1 328 108 A (UNILEVER LTD) 30 August 1973 cited in the application see page 1, line 65-92; claim 6	1
X	CA 1 255 603 A (COSMEPRO INC LAB) 13 June 1989 cited in the application see page 3, line 13 - page 4, line 9; claims 1-16	1-25
X	US 3 810 478 A (OLSON F ET AL) 14 May 1974 see column 5, line 46-52; example 2	1
X	EP 0 733 355 A (KAO CORP) 25 September 1996 see example 2	1-13, 15-25
X	GB 1 368 316 A (OREAL) 25 September 1974 see example 7	1
X	US 4 676 915 A (STELTENKAMP ROBERT J ET AL) 30 June 1987 see example 2	1
X	US 4 126 674 A (MAUSNER JACK J) 21 November 1978 see example 2	1-13, 15-25
1 X	EP 0 754 443 A (WELLA AG) 22 January 1997 see claims; examples 26-34	1-13, 15-25
1	-/-	

INTERNATIONAL SEARCH REPORT

Int'l Patent Application No
PCT/US 97/16410

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 94 06409 A (COLGATE PALMOLIVE CO) 31 March 1994 see examples 1,2 ---	1-13, 15-25
X	US 3 592 581 A (SHANSKY ALBERT ET AL) 13 July 1971 see column 9, line 60-70 ---	1-25
X	US 4 938 954 A (GROSS PAUL ET AL) 3 July 1990 see example 5 ---	1-25
X	WO 97 18795 A (OREAL ;AUDOUSSET MARIE PASCALE (FR); MONDET JEAN (FR)) 29 May 1997 see page 13, line 9-20; example 2 ---	1-25
X	DE 26 32 810 A (HENKEL KGAA) 26 January 1978 see examples ---	1-13, 15-25
X	EP 0 087 060 A (FABER CASTELL A W) 31 August 1983 see example 5 ---	1-25
A	GB 759 385 A (GILLETTE COMPANY) 17 October 1956 see page 2, line 30-70 ---	1-25
A	EP 0 336 709 A (DOW CORNING) 11 October 1989 see claims 2,5,20-22 -----	1-25

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No.

PCT/US 97/16410

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0717978 A	26-06-1996	US 5589177 A	31-12-1996
		AU 4026395 A	13-06-1996
		AU 688632 B	12-03-1998
		AU 4027195 A	13-06-1996
		CA 2163854 A	07-06-1996
		CN 1138982 A	01-01-1997
		FI 955866 A	07-06-1996
		NO 954916 A	07-06-1996
		NZ 280599 A	27-07-1997
		ZA 9510058 A	04-06-1996
GB 2307639 A	04-06-1997	DE 19646804 A	22-05-1997
		FR 2741261 A	23-05-1997
		JP 9183714 A	15-07-1997
US 3658985 A	25-04-1972	AT 302534 A	15-09-1972
		BE 753955 A	31-12-1970
		CH 531884 A	31-12-1972
		DE 2034295 A	11-02-1971
		DK 125620 B	19-03-1973
		FR 2053194 A	16-04-1971
		GB 1307644 A	21-02-1973
		NL 7011176 A	01-02-1971
		SE 351564 B	04-12-1972
		ZA 7004430 A	23-02-1972
GB 746864 A		BE 522974 A	
		FR 1094336 A	18-05-1955
US 4312855 A	26-01-1982	US 3875071 A	01-04-1975
		US 3832310 A	27-08-1974
		AR 196483 A	06-02-1974
		AT 322708 B	10-06-1975
		AU 3452671 A	19-04-1973
		BE 775158 A	01-03-1972
		CA 970909 A	15-07-1975
		CA 955390 A	01-10-1974
		DE 2155224 A	18-05-1972
		FR 2114751 A	30-06-1972
		GB 1357960 A	26-06-1974

INTERNATIONAL SEARCH REPORT

Information on patent family members

In International Application No

PCT/US 97/16410

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 4312855	A		NL 7115801 A SE 401608 B US 3703480 A US 3726815 A	18-05-1972 22-05-1978 21-11-1972 10-04-1973
EP 0834303	A	08-04-1998	DE 19640792 C DE 19640831 C	05-03-1998 05-03-1998
US 3577528	A	04-05-1971	DE 1617808 A GB 1152972 A NL 6703673 A	08-04-1971 21-05-1969 28-11-1967
GB 1328108	A	30-08-1973	DE 2064591 A	15-07-1971
CA 1255603	A	13-06-1989	NONE	
US 3810478	A	14-05-1974	CA 1020464 A GB 1414243 A	08-11-1977 19-11-1975
EP 0733355	A	25-09-1996	DE 19509981 A JP 8259426 A US 5635461 A	19-09-1996 08-10-1996 03-06-1997
GB 1368316	A	25-09-1974	AT 327395 B AU 471629 B AU 3470271 A BE 774082 A CA 968711 A CH 564347 A DE 2151789 A FR 2111698 A LU 61889 A NL 7114310 A US 3989447 A	26-01-1976 29-04-1976 03-05-1973 18-04-1972 03-06-1975 31-07-1975 20-04-1972 09-06-1972 28-06-1972 21-04-1972 02-11-1976
US 4676915	A	30-06-1987	IN 172493 A US 4682982 A US 4619775 A AT 395161 B AU 5847686 A	04-09-1993 28-07-1987 28-10-1986 12-10-1992 24-12-1986

INTERNATIONAL SEARCH REPORT

Information on patent family members

In International Application No

PCT/US 97/16410

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 4676915 A		BE 904932 A	16-12-1986
		BR 8602787 A	10-02-1987
		DE 3619807 A	18-12-1986
		DK 283386 A	18-12-1986
		EG 17896 A	30-10-1992
		FI 862521 A,B	18-12-1986
		FR 2583425 A	19-12-1986
		GB 2176814 A,B	07-01-1987
		IN 171622 A	28-11-1992
		IN 171860 A	23-01-1993
		JP 61291675 A	22-12-1986
		KR 9405274 B	15-06-1994
		LU 86475 A	13-01-1987
		NL 8601562 A	16-01-1987
		PT 82755 B	13-10-1987
		SE 8602608 A	18-12-1986
		AT 127186 A,B	15-07-1993
		AU 594337 B	08-03-1990
		AU 5723286 A	20-11-1986
		AU 616387 B	31-10-1991
		AU 8192287 A	09-06-1988
		BE 904782 A	17-11-1986
		BE 1000498 A	27-12-1988
		CA 1335206 A	11-04-1995
		CA 1284254 A	21-05-1991
		CH 677856 A,B	15-07-1991
		CH 675439 A	28-09-1990
		DE 3615500 A	20-11-1986
		DE 3740193 A	16-06-1988
		FR 2581992 A	21-11-1986
		FR 2609040 A	01-07-1988
		GB 2178028 A,B	04-02-1987
		GB 2198462 A,B	15-06-1988
		IE 59149 B	12-01-1994
		JP 61293285 A	24-12-1986
		JP 63170496 A	14-07-1988
		LU 86428 A	05-12-1986
		LU 87060 A	14-07-1988
		NL 8601258 A	16-12-1986
		NL 8702953 A	01-07-1988

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 97/16410

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 4676915 A		SE 8602188 A	17-11-1986
		SE 8704852 A	09-06-1988
		US 4764291 A	16-08-1988
		US 4714559 A	22-12-1987
		US 4715970 A	29-12-1987
US 4126674 A	21-11-1978	AR 218450 A	13-06-1980
		AU 508026 B	06-03-1980
		AU 2504877 A	16-11-1978
		BR 7702988 A	20-12-1977
		CA 1091585 A	16-12-1980
		DE 2721278 A	24-11-1977
		FR 2350835 A	09-12-1977
		JP 1391716 C	23-07-1987
		JP 53004010 A	14-01-1978
		JP 61054767 B	25-11-1986
		ZA 7702805 A	26-04-1978
EP 0754443 A	22-01-1997	DE 19525821 A	16-01-1997
		ES 2098208 T	01-05-1997
		JP 9030937 A	04-02-1997
WO 9406409 A	31-03-1994	AU 4928393 A	12-04-1994
		CN 1089830 A	27-07-1994
		MX 9305746 A	31-05-1994
		US 5580494 A	03-12-1996
		ZA 9306928 A	20-03-1995
US 3592581 A	13-07-1971	NONE	
US 4938954 A	03-07-1990	DE 3725080 A	09-02-1989
		DE 3864936 A	24-10-1991
		WO 8900845 A	09-02-1989
		EP 0301197 A	01-02-1989
		JP 2515149 B	10-07-1996
		JP 2500105 T	18-01-1990
WO 9718795 A	29-05-1997	FR 2741530 A	30-05-1997
		CA 2211797 A	29-05-1997
		EP 0808150 A	26-11-1997

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 97/16410

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9718795 A		JP 10502945 T	17-03-1998
DE 2632810 A	26-01-1978	NONE	
EP 0087060 A	31-08-1983	DE 3204636 A	25-08-1983
		JP 58198410 A	18-11-1983
GB 759385 A		NONE	
EP 0336709 A	11-10-1989	US 4866152 A	12-09-1989
		CA 1337524 A	07-11-1995
		JP 2011662 A	16-01-1990
		JP 2542077 B	09-10-1996
		US 4982603 A	08-01-1991
		US 4962177 A	09-10-1990